

IEAGHG

Overview and recent CCS activities

Jasmin Kemper

IEA Greenhouse Gas R&D Programme
Cheltenham, UK

MATESA Dissemination Day

16 June 2016, Oslo

Greenhouse Gas R&D TCP



Part of the IEA Energy Technology Network since 1991

35 Members from 18 countries plus OPEC, EU and CIAB

Members set strategic direction and technical programme

Universally recognised as independent technical organisation



What we are:



Current membership



ieaghg



Partner Organisations:



What do we do?



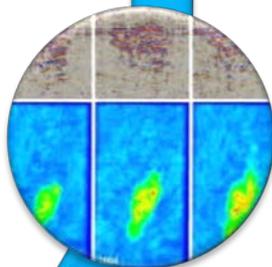
Our core activities are:



Assess mitigation options –
Focus our R&D on CCS



Facilitate technology implementation



Facilitate international co-operation



Disseminate our results as widely as possible

Technical studies



Technical and economic evaluations of technology options with the potential to mitigate GHG emissions

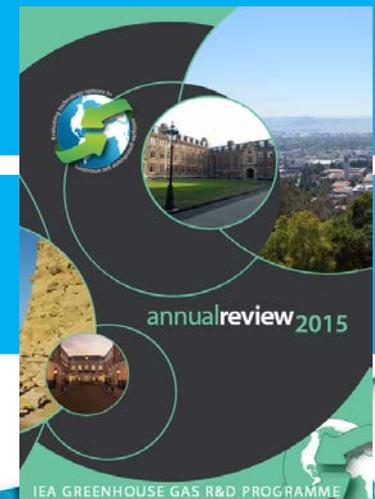
Available to individuals/organisations in all member countries and to all sponsor organisations upon publication

Available to those from non-member countries after a six month period

>250 in total on all aspects of CCS

12 – 15 technical reports each year

http://www.ieaghg.org/docs/General_Docs/Publications/Annual_Review_2015_Low_Res.pdf



Other dissemination activities



- Information papers (IPs)
- Blog
- Newsletter (weekly, quarterly)
- Webinars
- International Journal of Greenhouse Gas Control (IJGGC)

Latest
from
our

blog

10/06/2016 Hitting new highs and lows and raised concerns

The US National Oceanographic and Atmospheric Administration (NOAA)...

09/06/2016 Public Sharing of Information on Progressing Development of the UK's Strategic Carbon Dioxide Storage Resource

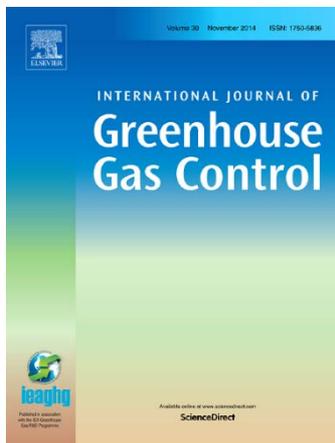
There was an excellent webinar today hosted by the Global CCS Institute on work from the UK's Energy Technologies Institute...

07/06/2016 Review of project permits under the London Protocol – An assessment of the proposed P18-4 CO₂ storage site

The London Convention and Protocol promotes the protection of the marine...

Full blog here

<http://www.ieaghq.org/publications/blog>



<http://www.journals.elsevier.com/international-journal-of-greenhouse-gas-control/>



<http://www.ieaghq.org/publications/greenhouse-news>

Networking activities



- 7 international research networks
- Conference series:
 - Greenhouse Gas Control Technologies Conference Series (GHGT)
 - Post-Combustion Capture Conference (PCCC)
 - Oxyfuel Combustion Conference (OCC)
- Summer School



High temperature solid looping cycles network (HTSLCN)



Covering the following topics:

- Calcium and chemical looping
- Combustion / gasification / reforming
- Fundamentals / modelling / testing

Constantly >50 attendees, focus on academia

Moving to a 2-year format to align with the International Conference on Chemical Looping

Next meeting 4-5 September 2017 in Luleå, Sweden

2009



Oviedo

2010



Petten

2011



Vienna

2012



Beijing

2013



Cambridge

2015



Milan

Emerging CO₂ capture technologies



Identify and review the main emerging capture technologies being developed for power plants

- Post-combustion capture
- Pre-combustion capture
- Oxy-combustion
- Solid looping

Assess current status and Technology Readiness Level (TRL)

Critically assess claims for energy requirements and cost reductions

Capture in non-power industries considered in less detail

Study did not involve detailed assessment of energy requirements and costs of plants with CO₂ capture



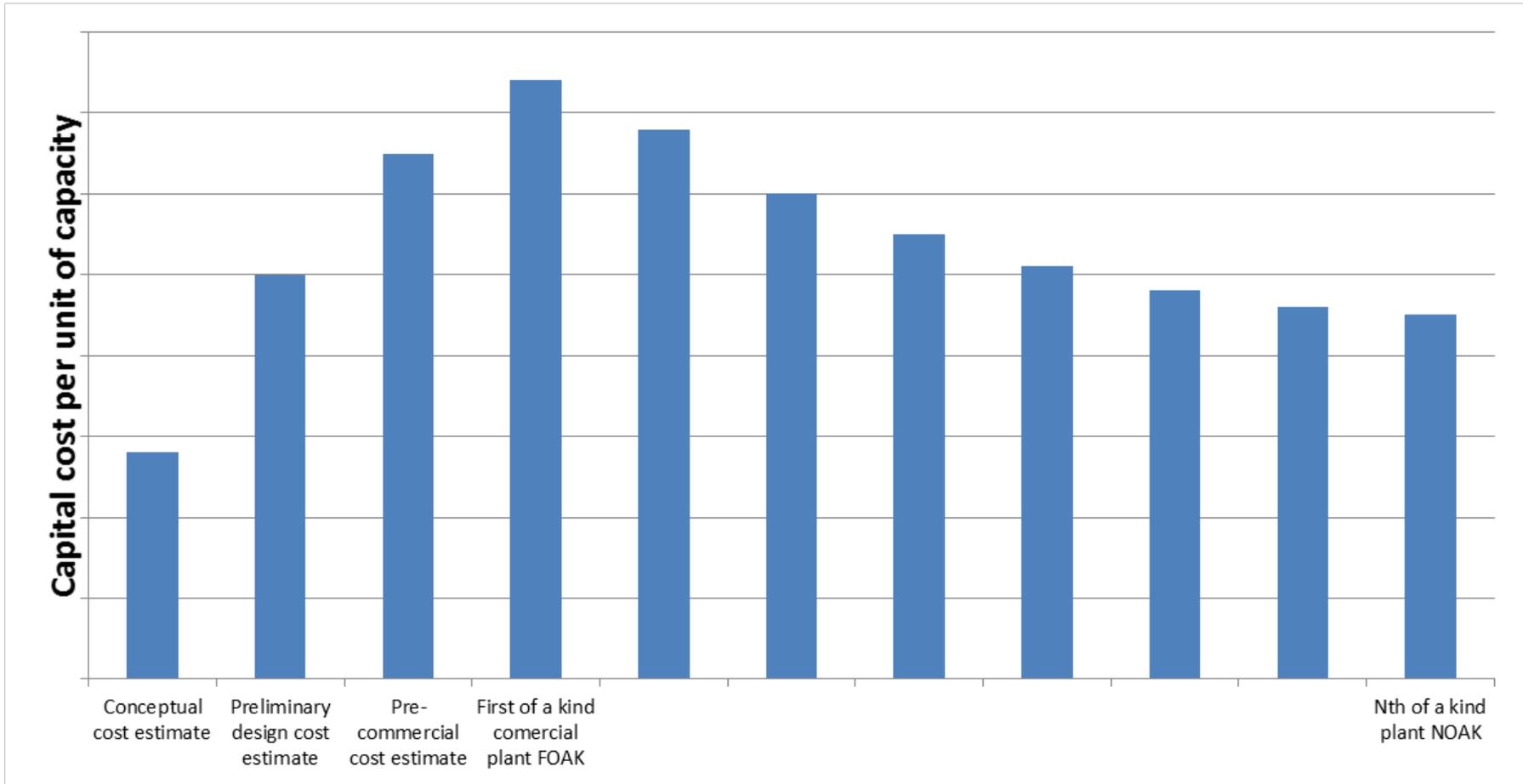
ASSESSMENT OF
EMERGING CO₂ CAPTURE
TECHNOLOGIES AND
THEIR POTENTIAL TO
REDUCE COSTS

Report: 2014/TR4

December 2014

http://www.ieaghg.org/docs/General_Docs/Reports/2014-TR4.pdf

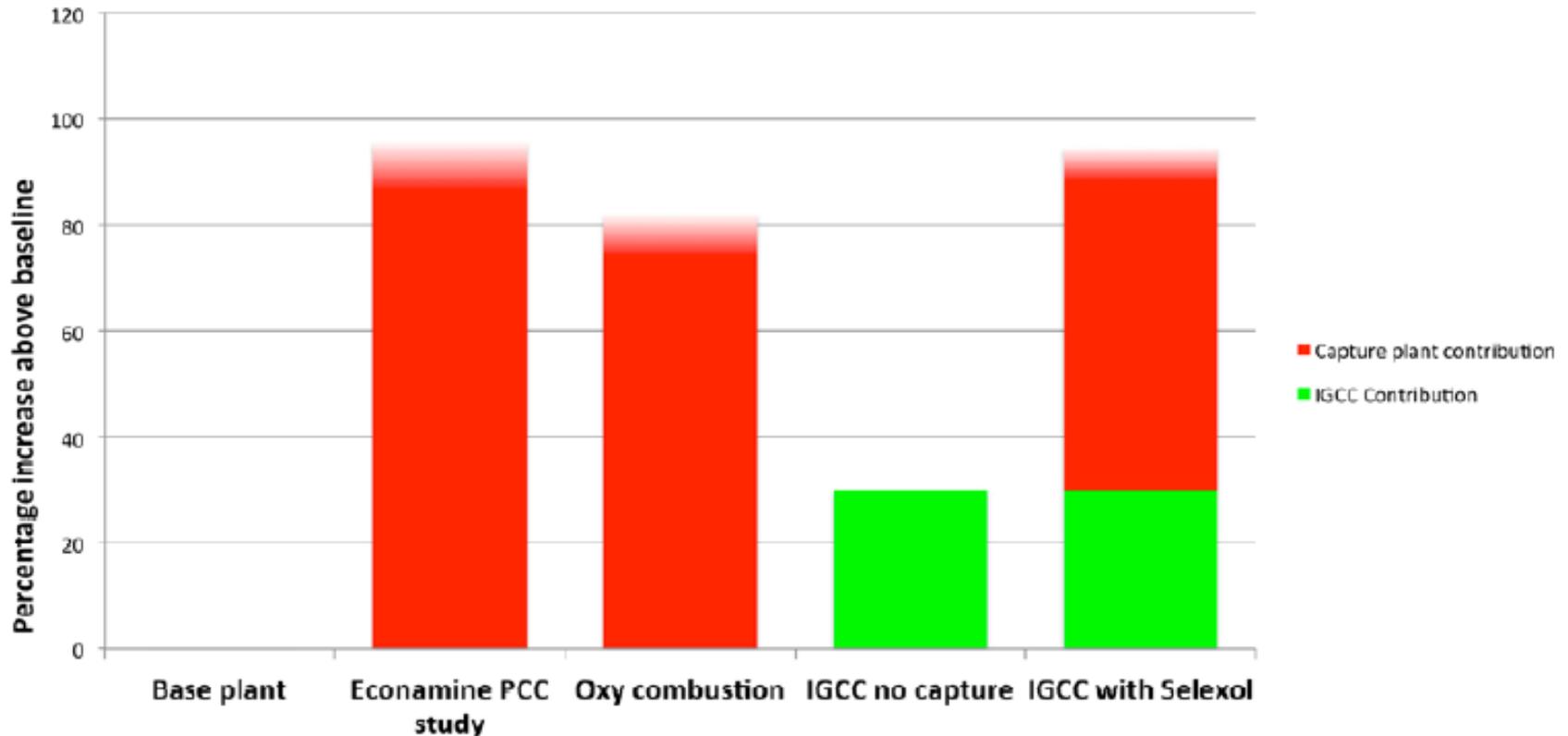
Cost learning curve



LCOE for CO₂ capture technologies



**Estimated percentage increases in LCOE
due to addition of CO₂ capture**
Benchmark post, oxy and pre combustion capture
Supercritical steam, coal fired power plant as baseline



Drivers for cost of capture



Capital cost of capture equipment

- Capital charges, cost of maintenance etc.

Increased fuel consumption

Increased specific capital cost of the host power generation process due to increased fuel consumption

Increased variable operating costs

- Capture solvent make-up etc.

→ Early stage assessments tend to focus initially on energy consumption

- Can be evaluated more scientifically
- A major contribution to capture cost

Post-combustion capture



TRL 1 - 3

- Enzyme catalysed adsorption
- Ionic liquids
- Room temperature ionic liquid (RTIL) membranes
- Encapsulated solvents
- Electrochemically mediated absorption
- Vacuum pressure swing adsorption (VPSA)
- Cryogenic capture
- Supersonic inertial capture

TRL 4 – 6

- Bi-phasic solvents
- Precipitating solvents
- Polymeric membranes
- Temperature swing adsorption

TRL 7 – 9

- Benchmark amine scrubbing
- Improved conventional solvents

Solid looping processes



TRL 1 - 3

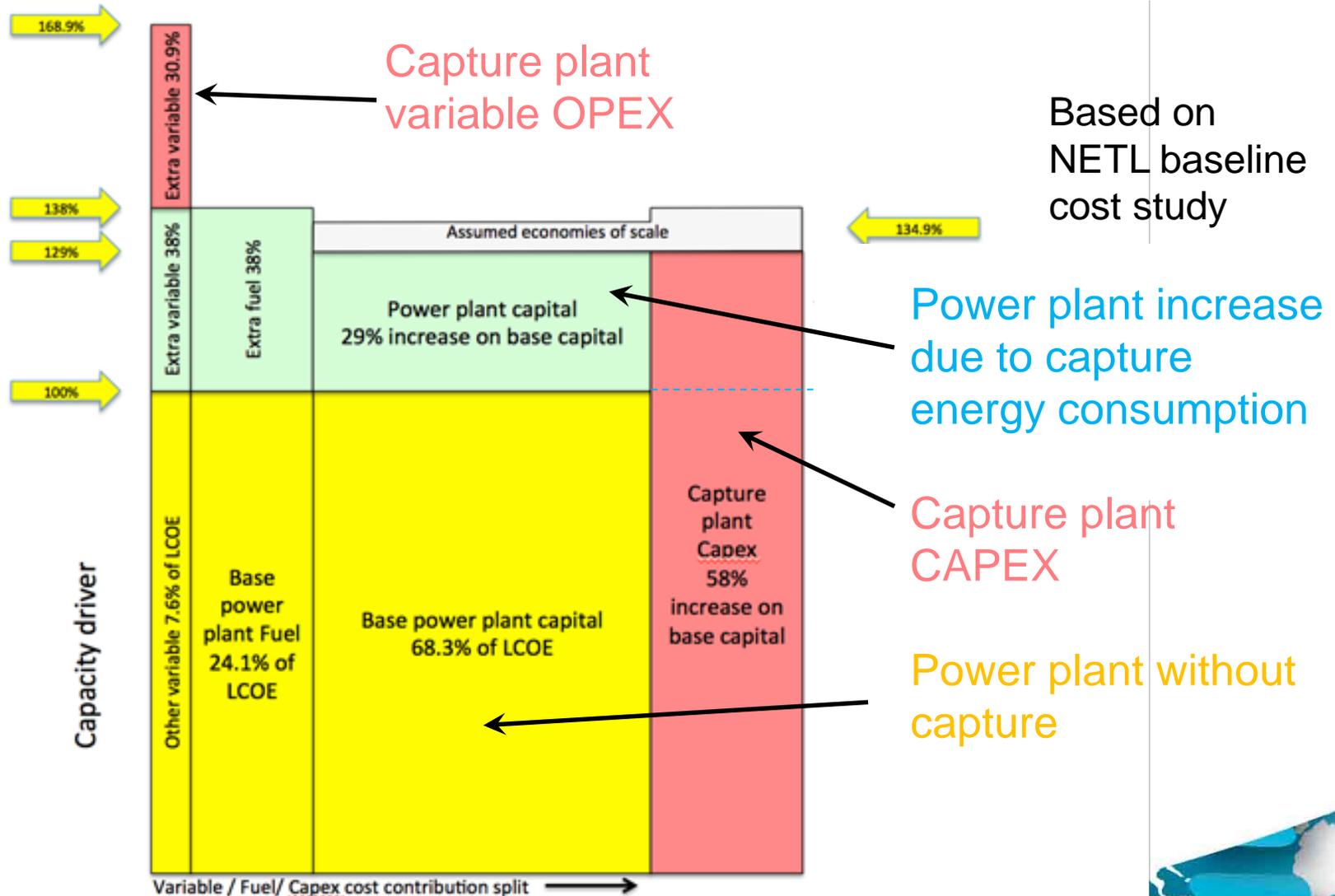
- Sorption enhanced reforming (SER)
- Chemical looping gasification (CLG)
- Chemical looping with oxygen uncoupling (CLOU)
- etc.

TRL 4 – 6

- Calcium carbonate looping (CaL)
- Chemical looping combustion (CLC)

TRL 7 - 9

Post-combustion capture Contributions to cost of electricity

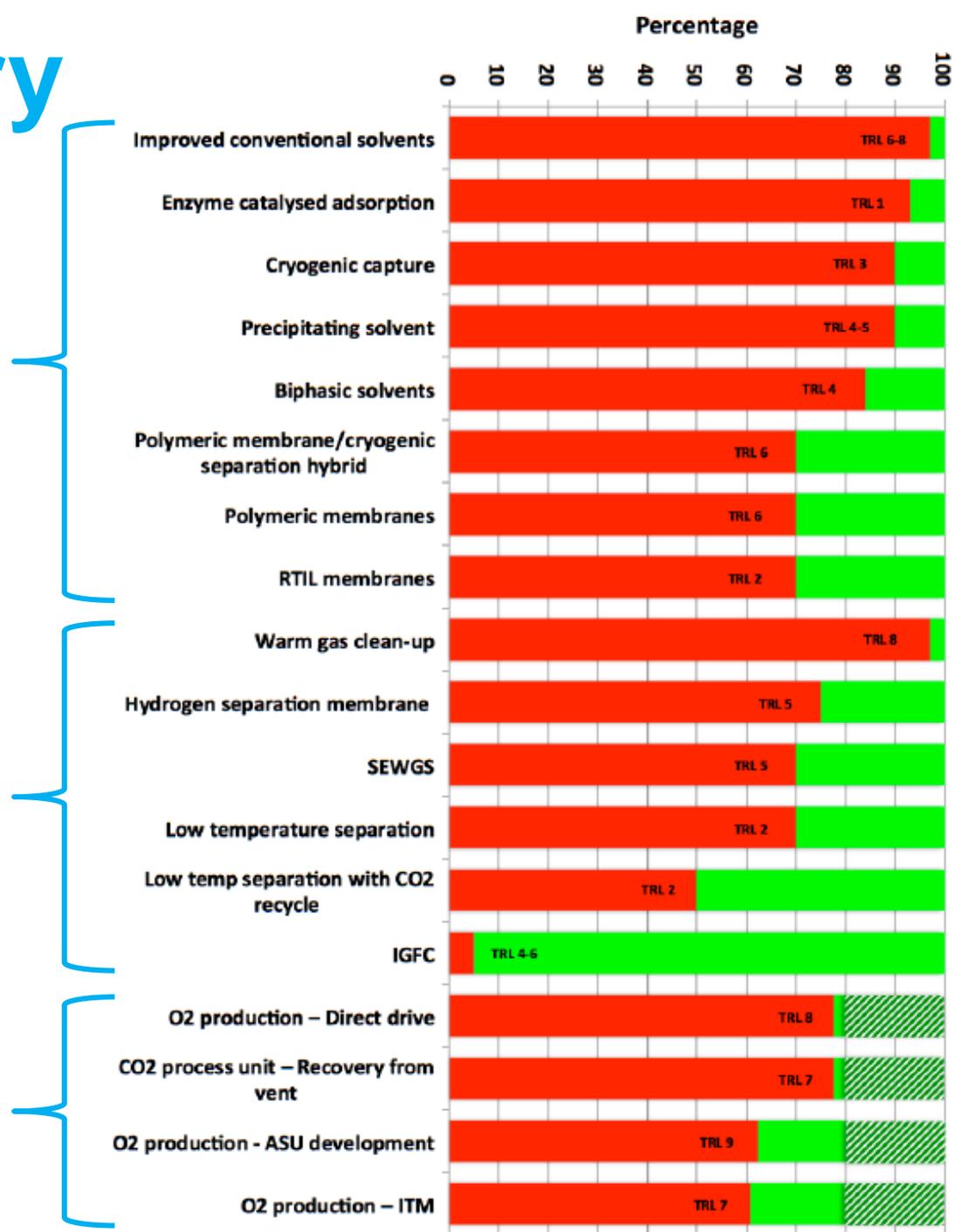


Summary

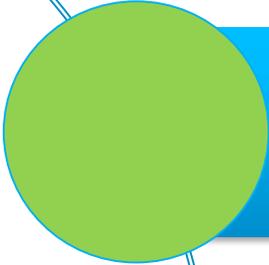
Post-combustion capture

Pre-combustion capture

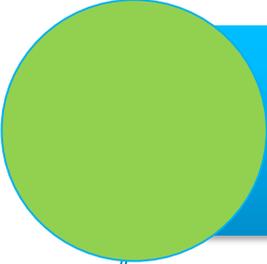
Oxy-combustion capture



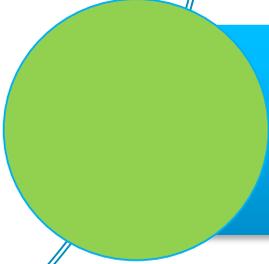
Conclusions



Many new technologies for CO₂ capture are being developed



Estimated costs of new capture technologies are subject to high uncertainty, especially at low TRLs



Processes in which CO₂ capture is a more integrated part of the power generation process show high potential for energy and cost reduction but have significant development hurdles

- E.g. solid looping combustion, oxy-combustion turbines and fuel cells
- 
- A large, stylized globe graphic at the bottom of the slide, showing continents in white and oceans in blue.

CO₂ capture in natural gas production by adsorption processes



■ Main objectives:



1. Evaluate utilisation of PSA process for CO₂ removal from NG
 2. Perform techno-economic comparison of PSA with a reference process, i.e. solvent scrubbing
 3. Investigate candidate materials for kinetic adsorbents
 4. Provide recommendations for future work
- PSA unit design will not include final and detailed process optimisation
 - Innovation of this work:
 - Novel process design not reported in literature so far

Existing technologies



- Hollow fibre modules
- Flat sheet spiral modules
- + straightforward, low energy and footprint
- - pre-treatment, fouling

Membrane

- + well known, deliver of NG at pipeline spec, flexibility
- - avoid freezing below triple point, azeotrope, costs
- Suitable for LNG

Cryogenic

- Temperature swing
- Pressure swing
- + lower energy demand and OPEX
- - Limited materials, CO₂ removal in R&D

Adsorption

- + well known, suitable for range of conditions, removes CO₂ and H₂S to ppm
- - High energy demand, degradation

Solvents



Conditions



Raw NG conditions and composition

Temperature [°C]	40
Pressure [bar]	70
CH ₄ [vol%]	83
C ₂ H ₆ [vol%]	4.6
C ₃₊ [vol%]	2.4
CO ₂ [vol%]	10

Sweet NG specifications

Temperature [°C]	40
Pressure [bar]	70
Lower heating value (LHV) [MJ/kg]	39
CO ₂ content [mol%]	≤ 2.5

CO₂ stream specifications

Temperature [°C]	40
Pressure [bar]	110
CO ₂ purity [vol%]	≥ 95

3 cost KPIs:

- 1) NG sweetening
- 2) CO₂ removal w/ and w/o CO₂ conditioning, transport and storage
- 3) CO₂ avoidance

Reference case: aMDEA



- Chemical solvent based NG upgrading process modelled with ProTreat v4.2
- 45wt% MDEA + 5wt% PZ (aMDEA)
- Regeneration mainly by pressure release
- Temperature of lean solvent feed to absorber is set $>10^{\circ}\text{C}$ higher than dew point of sweet gas
 - Avoid co-adsorption of potential heavy hydrocarbons

PSA process - adsorbent



- Adsorbent selection is the main and initial task in the specification of a PSA unit
- A direct reliable method of selecting adsorbents is currently not available → experience
- Two issues influence selection:
 1. Non-linear isotherms for CO_2
 2. Co-adsorption of CH_4
- To limit adsorption of CH_4 → kinetic adsorbents
 - a. Titanosilicates → commercially available, samples not
 - b. Carbon molecular sieve (CMS) → readily available, so used here

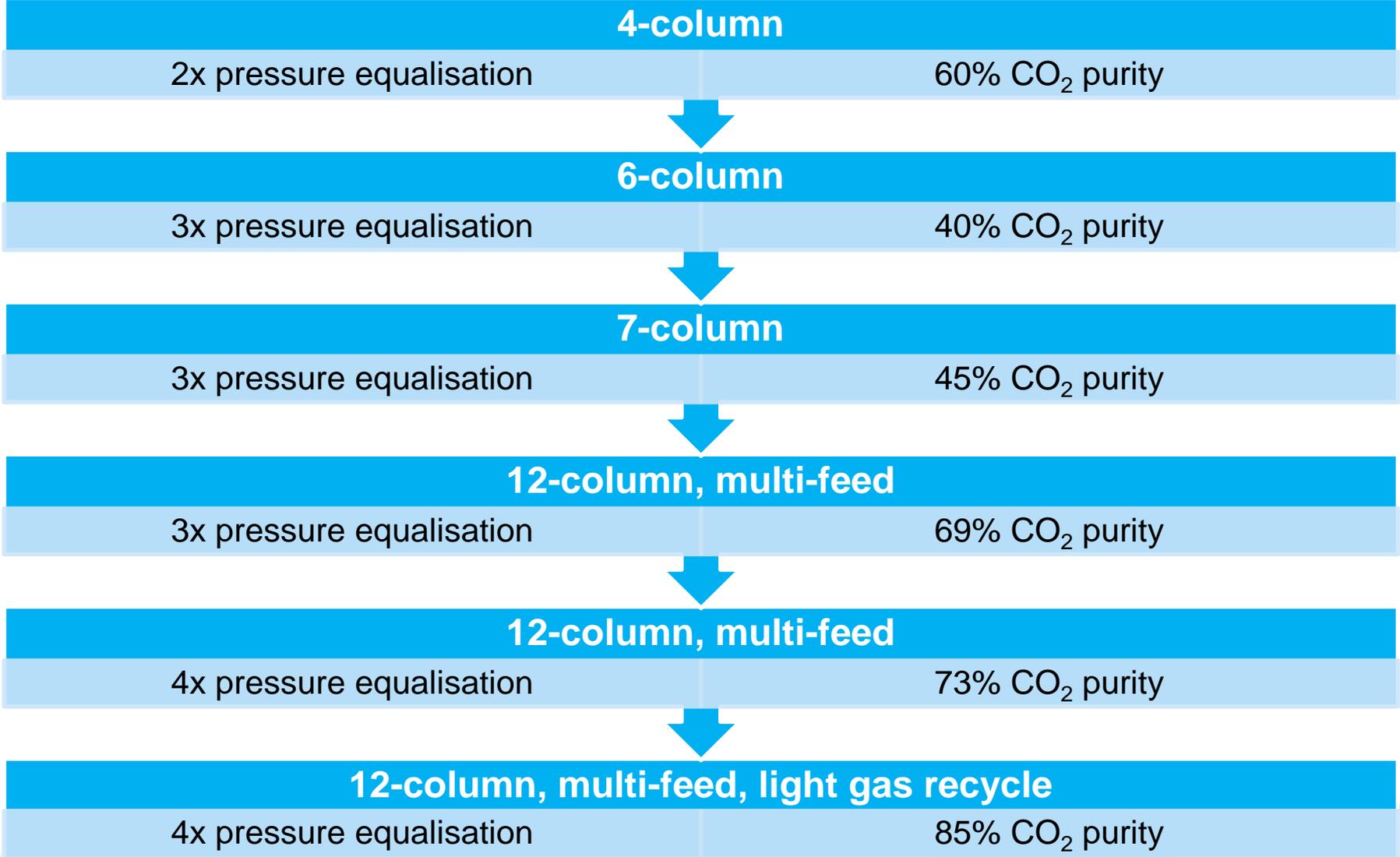
PSA process - model



- Several commercial programmes available
- This study used gPROMS
- Two approaches to PSA modelling
 1. Simulate performance of entire PSA by solving a model for only one column
 - Straightforward
 - Limited accuracy
 2. Simulate performance of PSA with a dynamic model of the whole system
 - Very detailed
 - High computation time (up to 20h for one pass)
 - Providing the right initial conditions is critical for convergence due to strong variation of conditions in a PSA



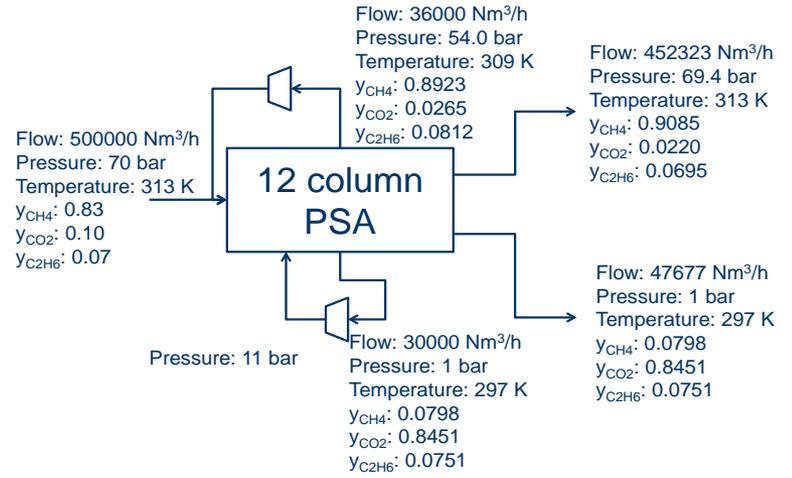
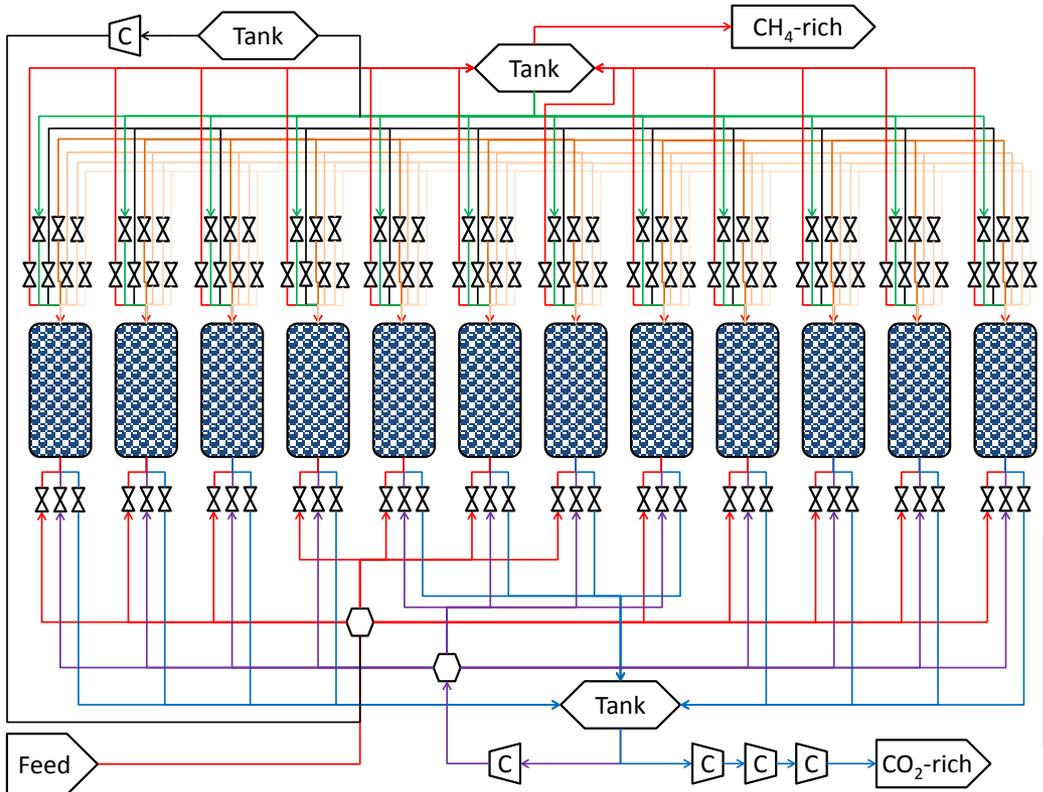
PSA process – iterative cycle design



Target: 95%



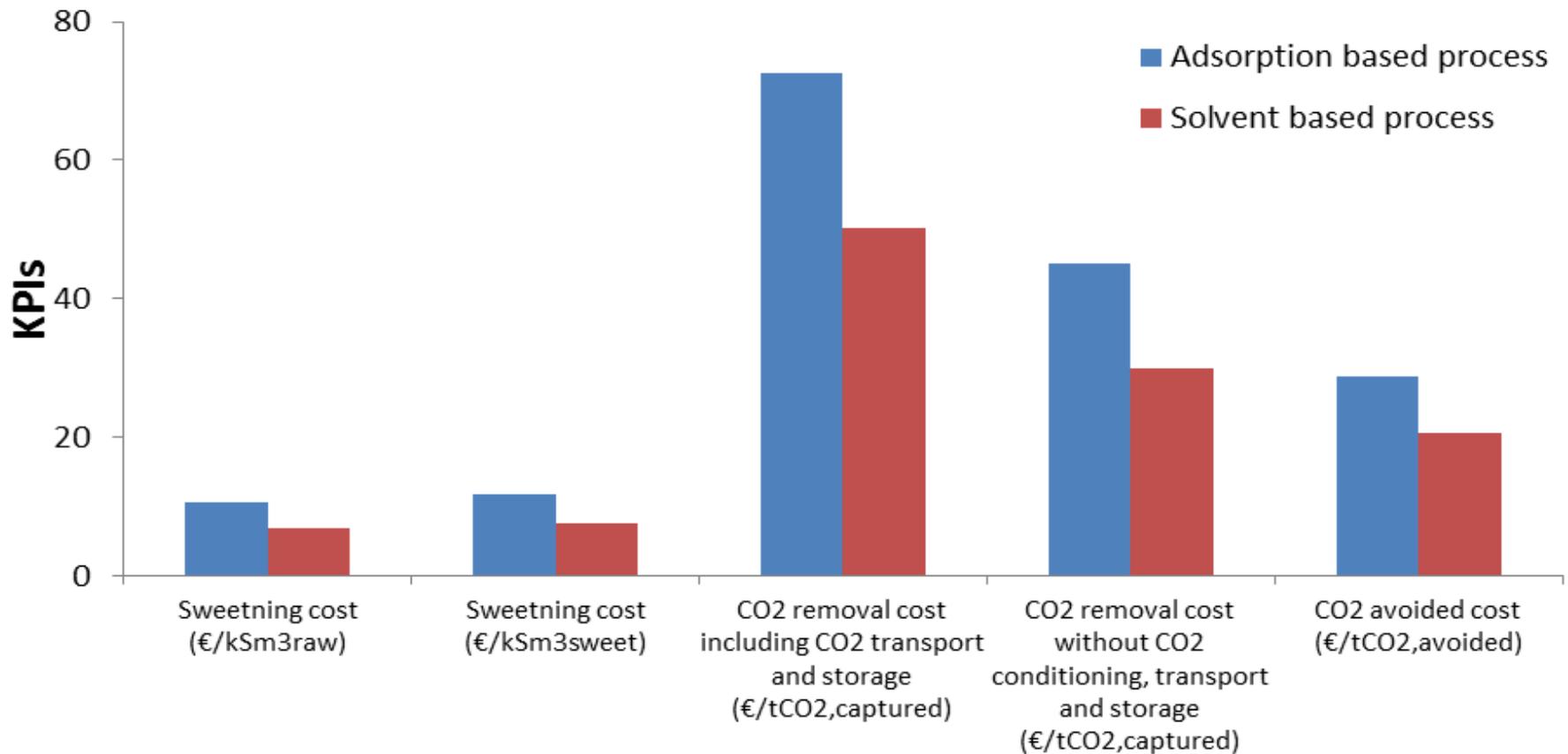
PSA process – final cycle design



C1			FEED ↑		D1 ↑	D2 ↑	D3 ↓	D4 ↑	PP ↑	R ↑	B ↓	Pu ↓	Pu ↓	E4 ↓	E3 ↓	E2 ↓	E1 ↓	Pr ↓	
C2	E1 ↓	Pr ↓		FEED ↑		D1 ↑	D2 ↑	D3 ↓	D4 ↑	PP ↑	R ↑	B ↓	Pu ↓	Pu ↓	E4 ↓	E3 ↓	E2 ↓	E1 ↓	
C3	E3 ↓	E2 ↓	E1 ↓	Pr ↓		FEED ↑		D1 ↑	D2 ↑	D3 ↓	D4 ↑	PP ↑	R ↑	B ↓	Pu ↓	Pu ↓	E4 ↓	E3 ↓	
C4	Pu ↓	E4 ↓	E3 ↓	E2 ↓	E1 ↓	Pr ↓		FEED ↑		D1 ↑	D2 ↑	D3 ↓	D4 ↑	PP ↑	R ↑	B ↓	Pu ↓	Pu ↓	
C5	Pu ↓	Pu ↓	E4 ↓	E3 ↓	E2 ↓	E1 ↓	Pr ↓		FEED ↑		D1 ↑	D2 ↑	D3 ↓	D4 ↑	PP ↑	R ↑	B ↓	Pu ↓	
C6	B ↓	Pu ↓	Pu ↓	E4 ↓	E3 ↓	E2 ↓	E1 ↓	Pr ↓		FEED ↑		D1 ↑	D2 ↑	D3 ↓	D4 ↑	PP ↑	R ↑	B ↓	
C7	R ↑		B ↓	Pu ↓	Pu ↓	E4 ↓	E3 ↓	E2 ↓	E1 ↓	Pr ↓		FEED ↑		D1 ↑	D2 ↑	D3 ↓	D4 ↑	PP ↑	
C8	PP ↑	R ↑		B ↓	Pu ↓	Pu ↓	E4 ↓	E3 ↓	E2 ↓	E1 ↓	Pr ↓		FEED ↑		D1 ↑	D2 ↑	D3 ↓	D4 ↑	
C9	D3 ↑	D4 ↑		PP ↑	R ↑		B ↓	Pu ↓	Pu ↓	E4 ↓	E3 ↓	E2 ↓	E1 ↓	Pr ↓		FEED ↑		D1 ↑	
C10	D1 ↑	D2 ↑	D3 ↑	D4 ↑		PP ↑	R ↑		B ↓	Pu ↓	Pu ↓	E4 ↓	E3 ↓	E2 ↓	E1 ↓	Pr ↓		FEED ↑	
C11	FEED ↑	D1 ↑	D2 ↑	D3 ↑	D4 ↑		PP ↑	R ↑		B ↓	Pu ↓	Pu ↓	E4 ↓	E3 ↓	E2 ↓	E1 ↓	Pr ↓	FEED ↑	
C12		FEED ↑		D1 ↑	D2 ↑	D3 ↑	D4 ↑		PP ↑	R ↑		B ↓	Pu ↓	Pu ↓	E4 ↓	E3 ↓	E2 ↓	E1 ↓	Pr ↓

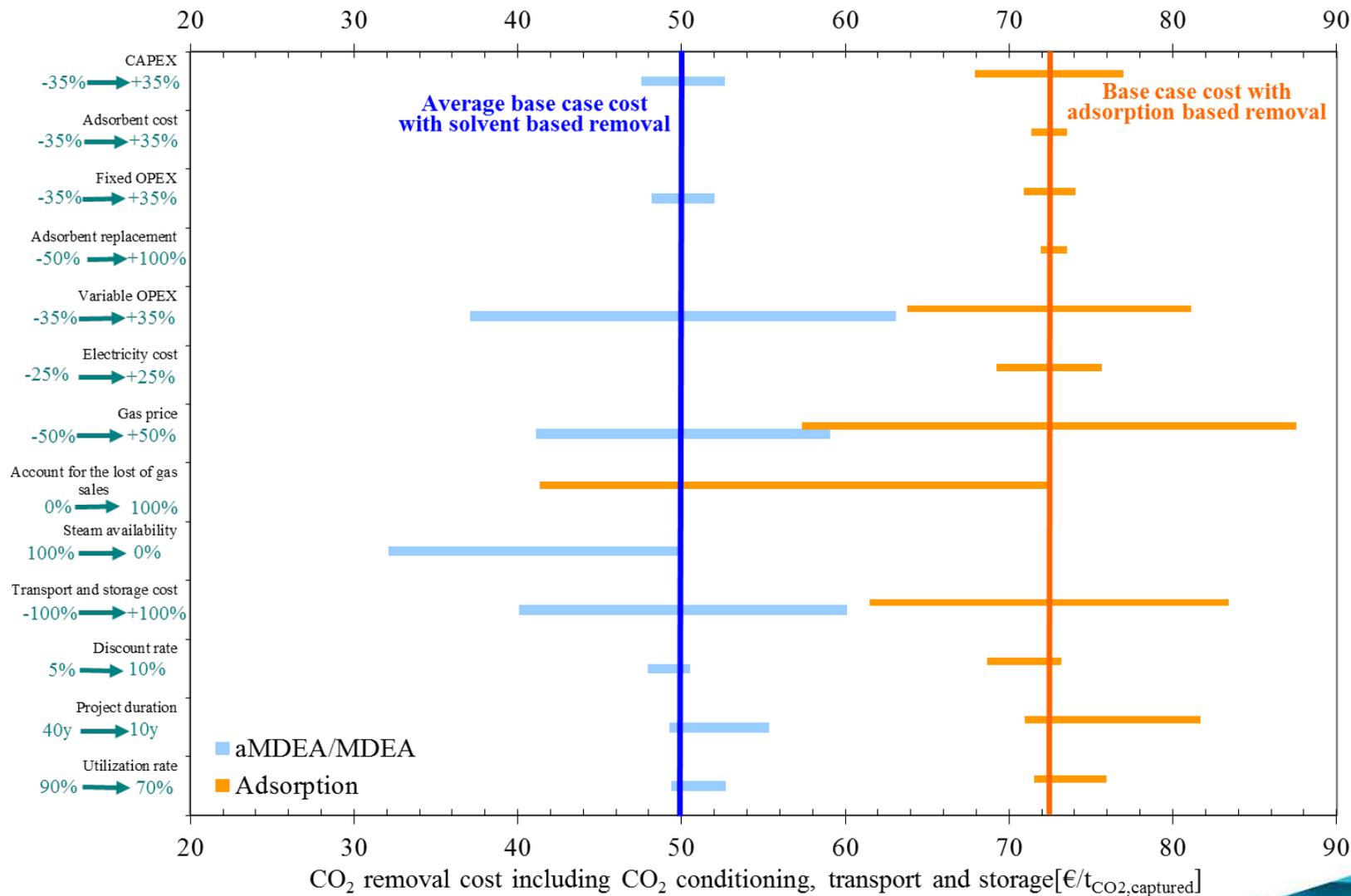
- Final 12-column multi-feed with four pressure equalisations and light gas recycle
- Process flow diagram with internal recycles and tanks

PSA – aMDEA cost comparison



- Differences between PSA and aMDEA mainly due to loss of NG

Cost sensitivity analysis



Conclusions



Iterative pathway was applied to find a PSA cycle design with maximum CO₂ purity

Final design is a 12-column multi-feed cycle with 85% CO₂ purity → first design for 70 bar and 500 000 Sm³/h

CO₂ removal and NG sweetening costs are ~50% higher than for the reference aMDEA amine process

Identified materials worth of future investigation

Process not yet optimised → ample room for improvement → combined approach of material and process optimisation can bring down cost significantly



Thank you, any questions?

Contact me at: jasmin.kemper@ieaghg.org



Website:

- www.ieaghg.org



LinkedIn:

- www.linkedin.com/groups/IEAGHG-4841998



Twitter:

- <https://twitter.com/IEAGHG>



Facebook:

- www.facebook.com/pages/IEA-Greenhouse-Gas-RD-Programme/112541615461568?ref=hl



ghgt-13



Registration opens 6 April 2016

Draft technical programme 1 June 2016

Early bird registration closes 13 July 2016



ÉCOLE POLYTECHNIQUE
FÉDÉRALE DE LAUSANNE



Schweizerische Eidgenossenschaft
Confédération suisse
Confederazione Svizzera
Confederaziun svizra

Swiss Confederation

Swiss Federal Office of Energy SFOE

www.ghgt.info



ieaghg

LAUSANNE, SWITZERLAND, NOVEMBER 14-18, 2016