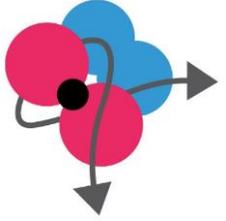




ELEGANCy

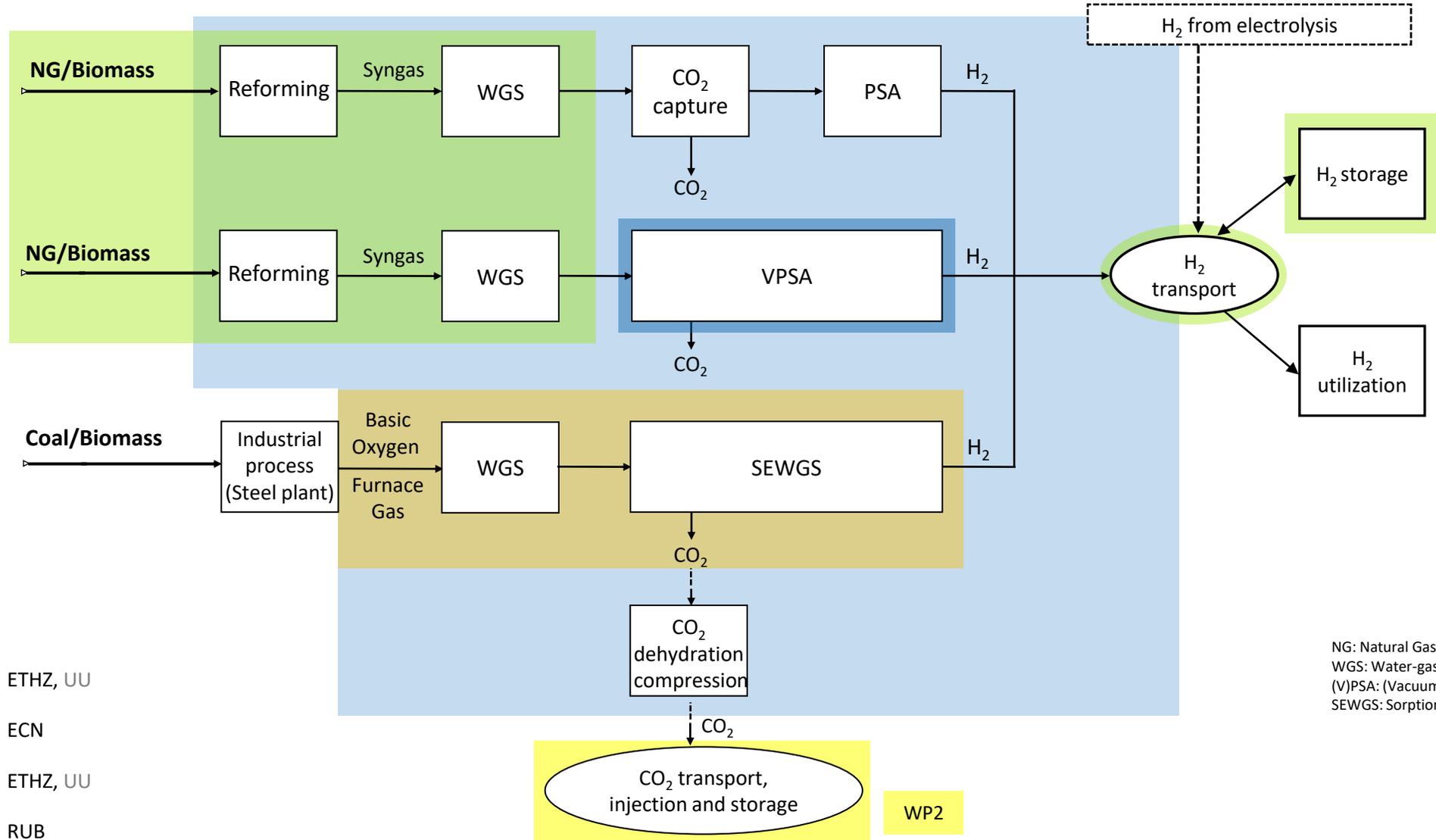
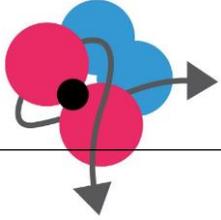


## Biomass to hydrogen with CCS: can we go negative?

Cristina Antonini and Marco Mazzotti

2020-06-22

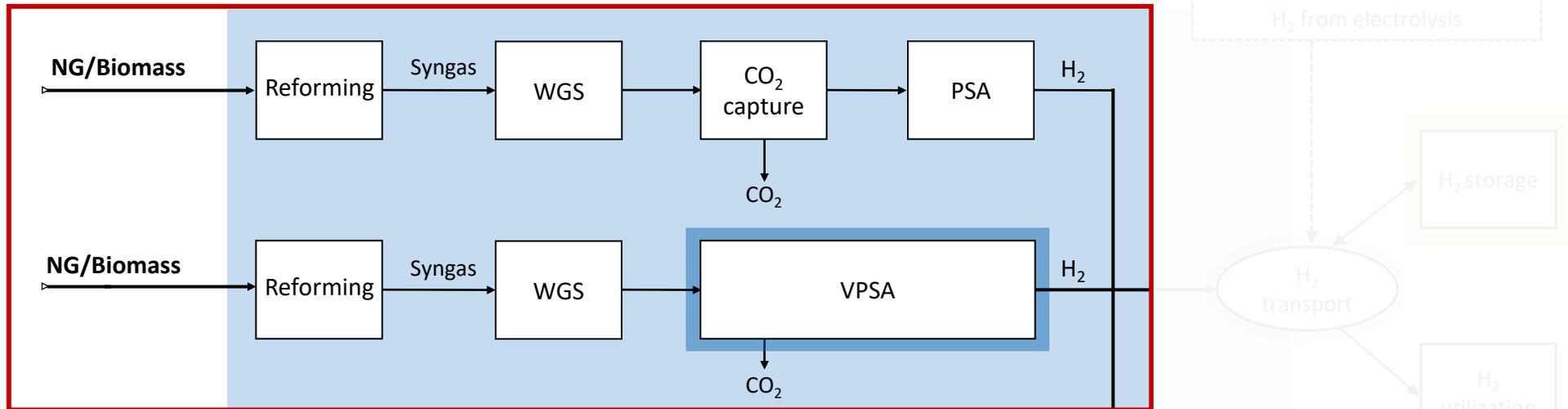
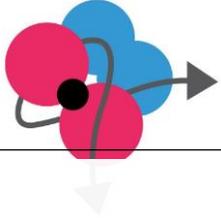
# ELEGANCY - Overview



- Task 1.1 ETHZ, UU
- Task 1.2 ECN
- Task 1.3 ETHZ, UU
- Task 1.4 RUB

NG: Natural Gas  
 WGS: Water-gas shift section  
 (V)PSA: (Vacuum) Pressure swing adsorption  
 SEWGS: Sorption enhanced WGS

# ELEGANCY - Low-C H<sub>2</sub> production



## Goals:

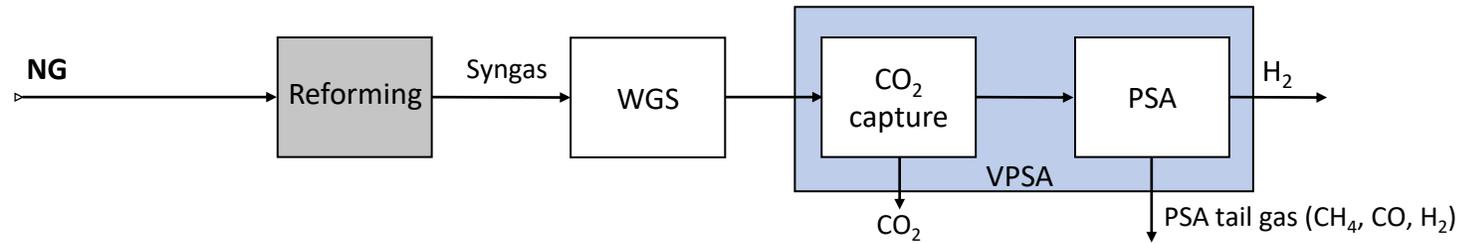
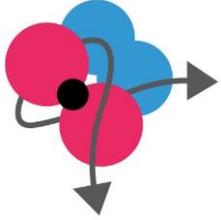
- 1) Study low-C hydrogen production with CO<sub>2</sub> capture and storage
  - starting from different feedstocks
  - using different production technologies
  - comparing benchmark with novel CO<sub>2</sub>/H<sub>2</sub> separation processes
- 2) Investigate the possibility to deliver negative emissions

NG: Natural Gas  
WGS: Water-gas shift section  
(V)PSA: (Vacuum) Pressure swing adsorption  
SEWGS: Sorption enhanced WGS

CO<sub>2</sub> transport  
injection and storage

WP2

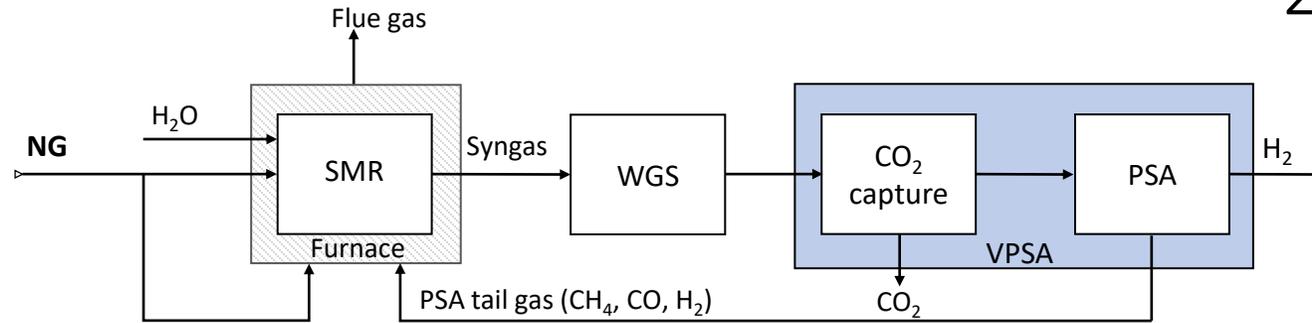
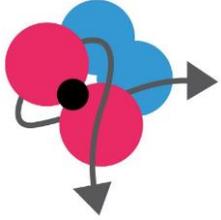
# Low-C H<sub>2</sub> production



Feedstock	Feedstock conversion	Technology
– Natural gas (NG)	Reforming	

NG: Natural Gas  
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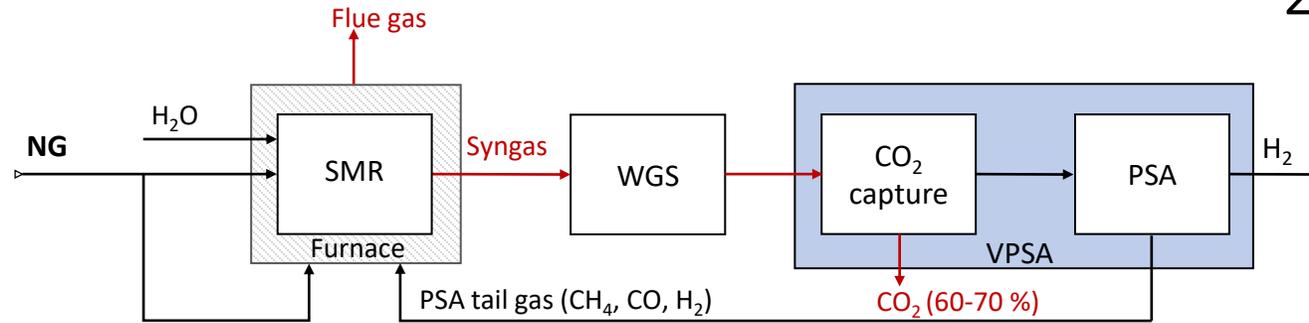
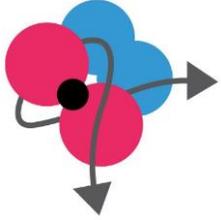
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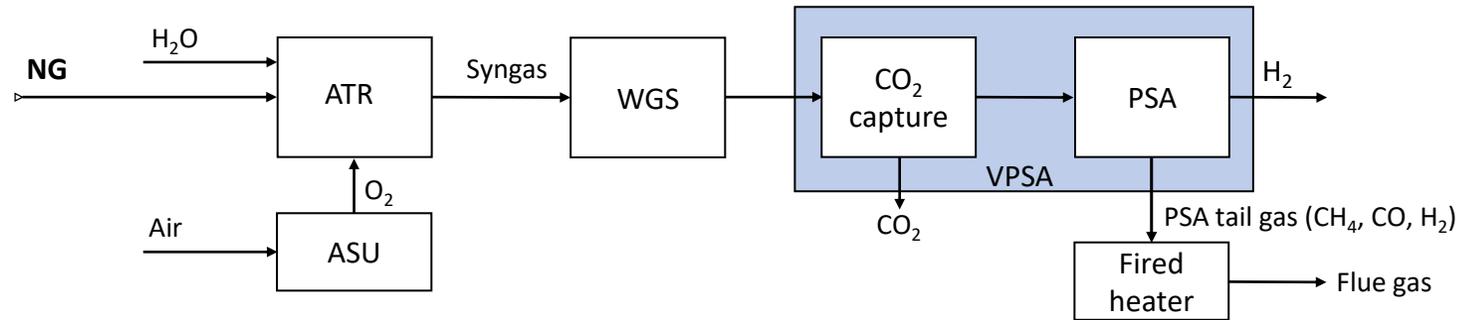
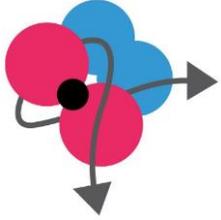


→ CO<sub>2</sub>

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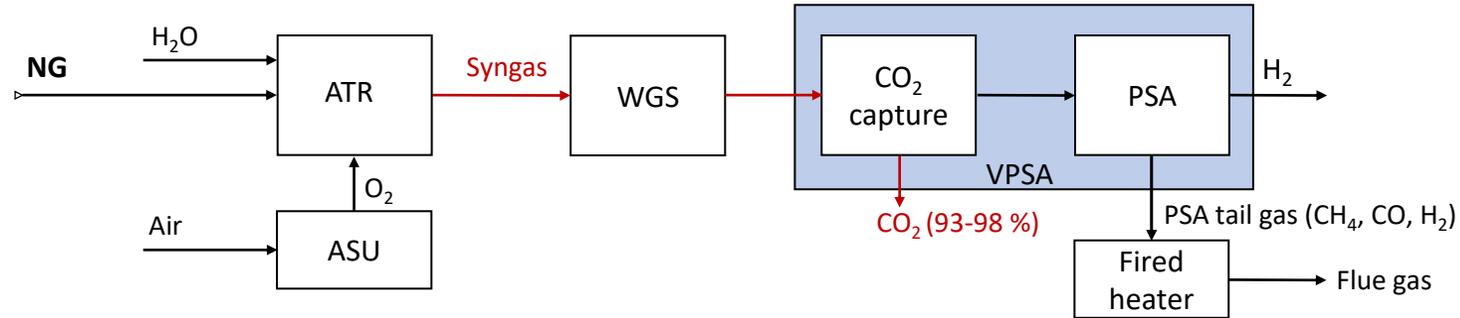
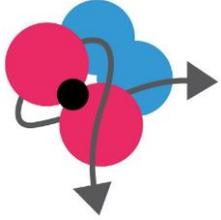
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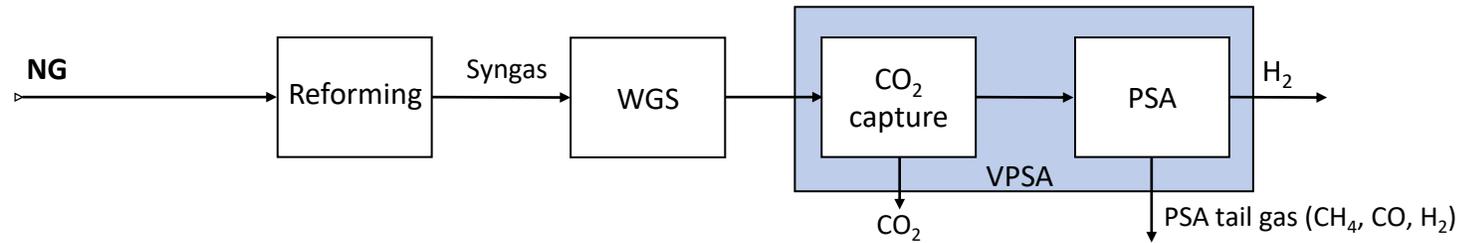
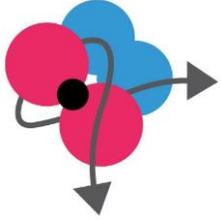


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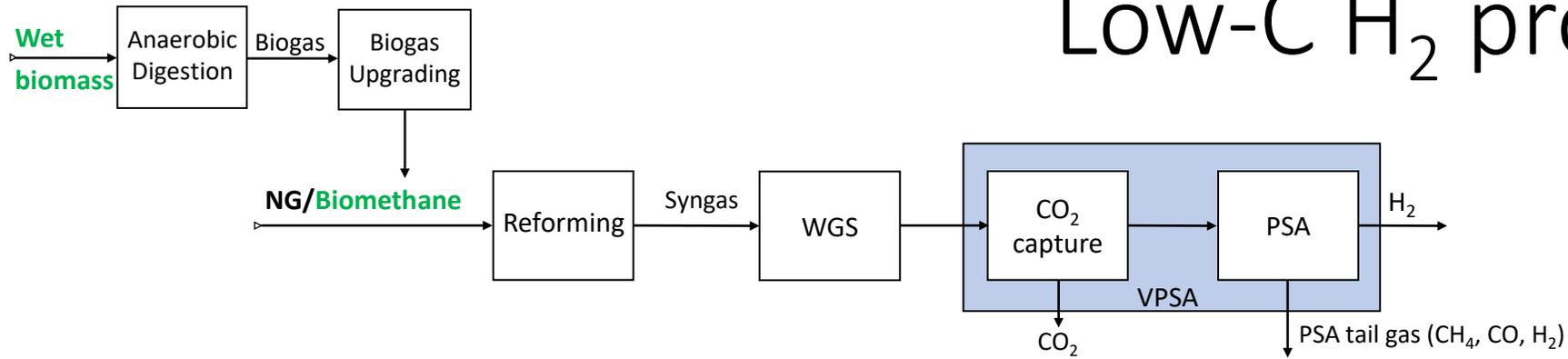
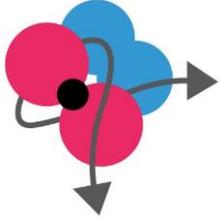
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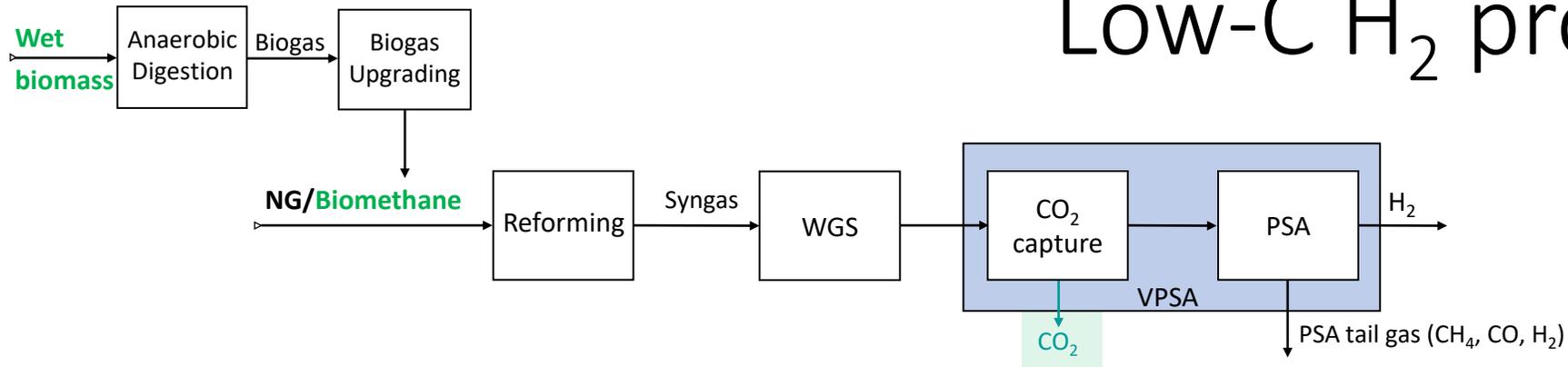
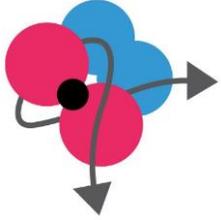
# Low-C H<sub>2</sub> production



Feedstock	Feedstock conversion	Technology
<ul style="list-style-type: none"> <li>Natural gas (NG)</li> <li>Biomethane from WB</li> </ul>	Reforming	<ul style="list-style-type: none"> <li>Steam methane reforming (SMR)</li> <li>Autothermal reforming (ATR)</li> </ul>

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# Low-C H<sub>2</sub> production

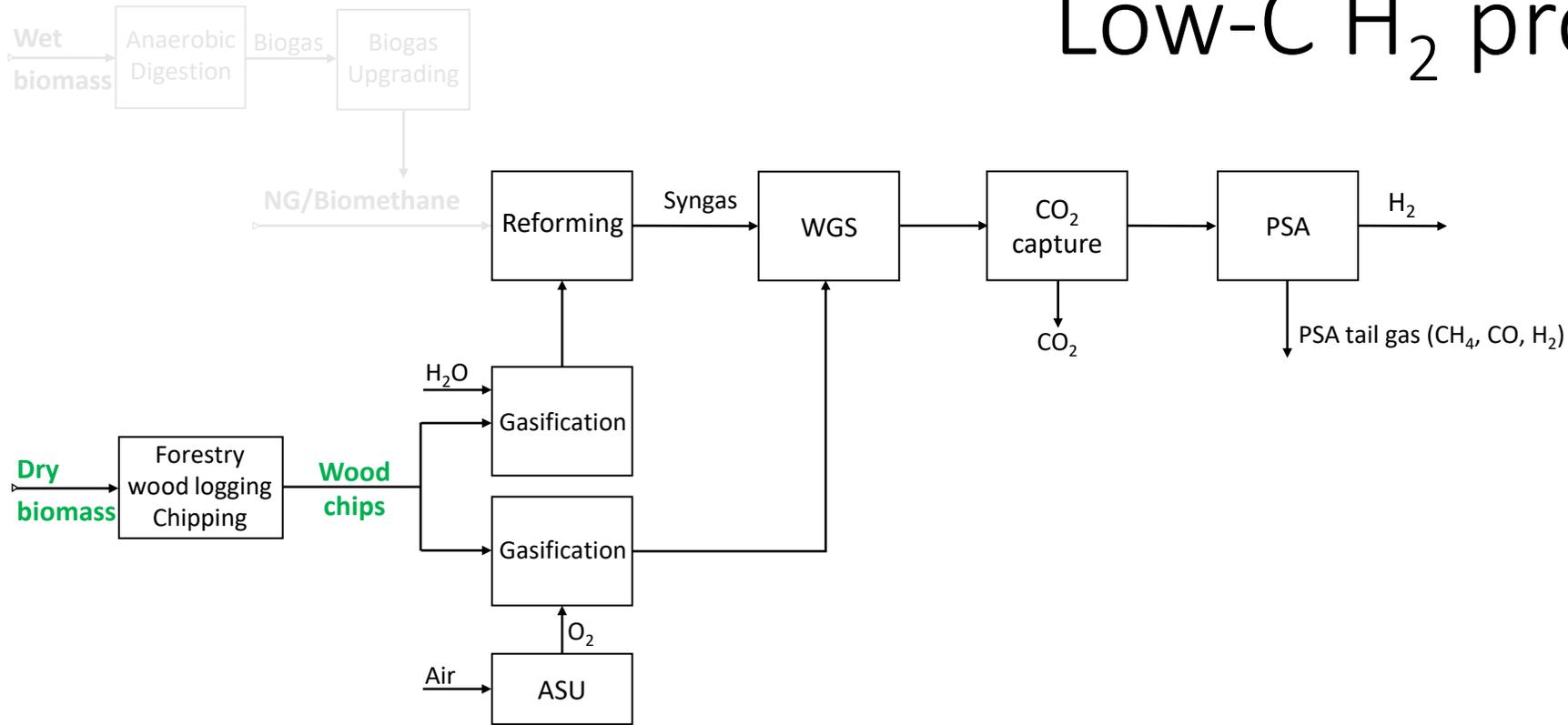
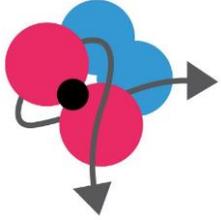


## Negative emissions

Feedstock	Feedstock conversion	Technology
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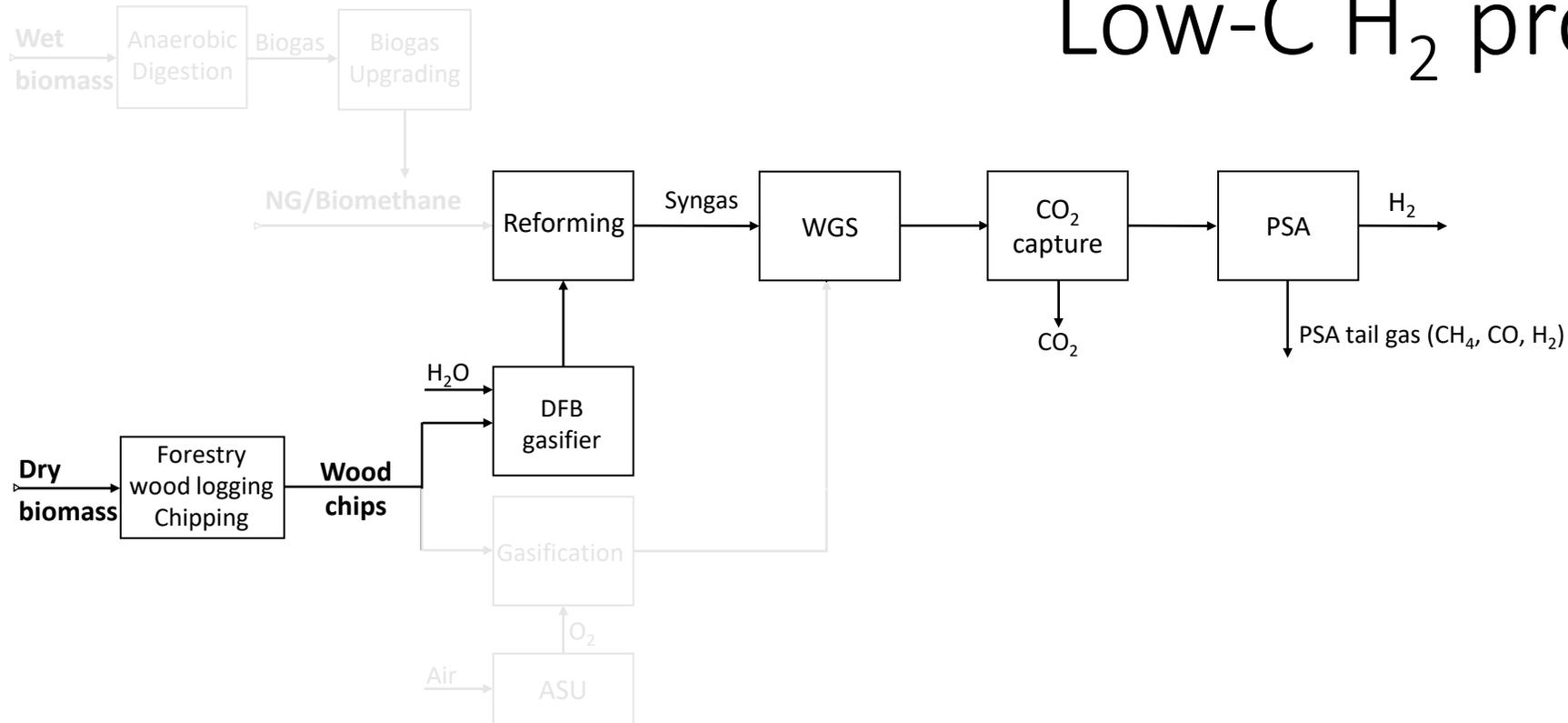
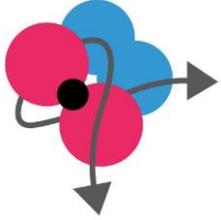
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Feedstock	Feedstock conversion	Technology
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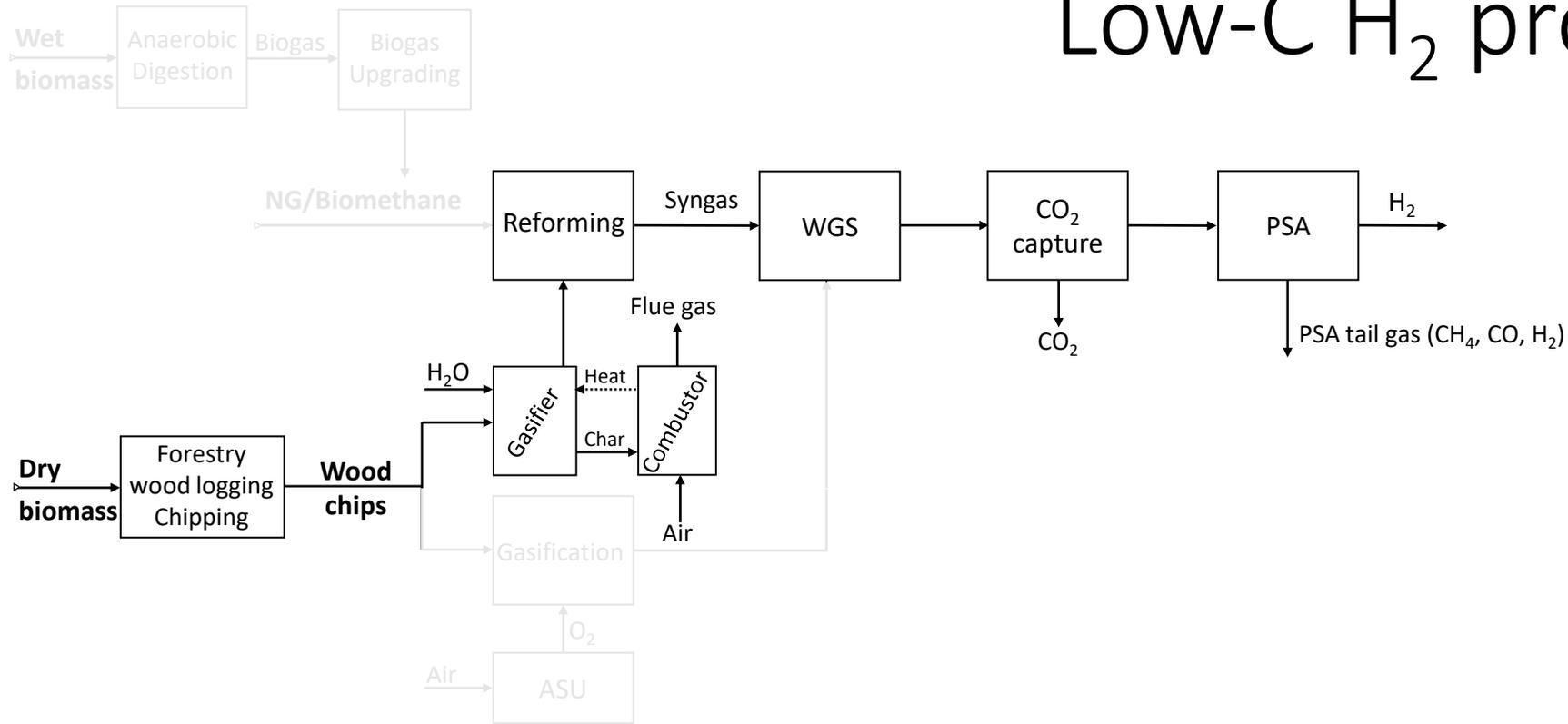
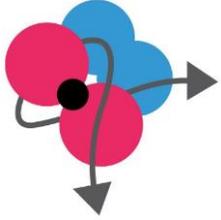
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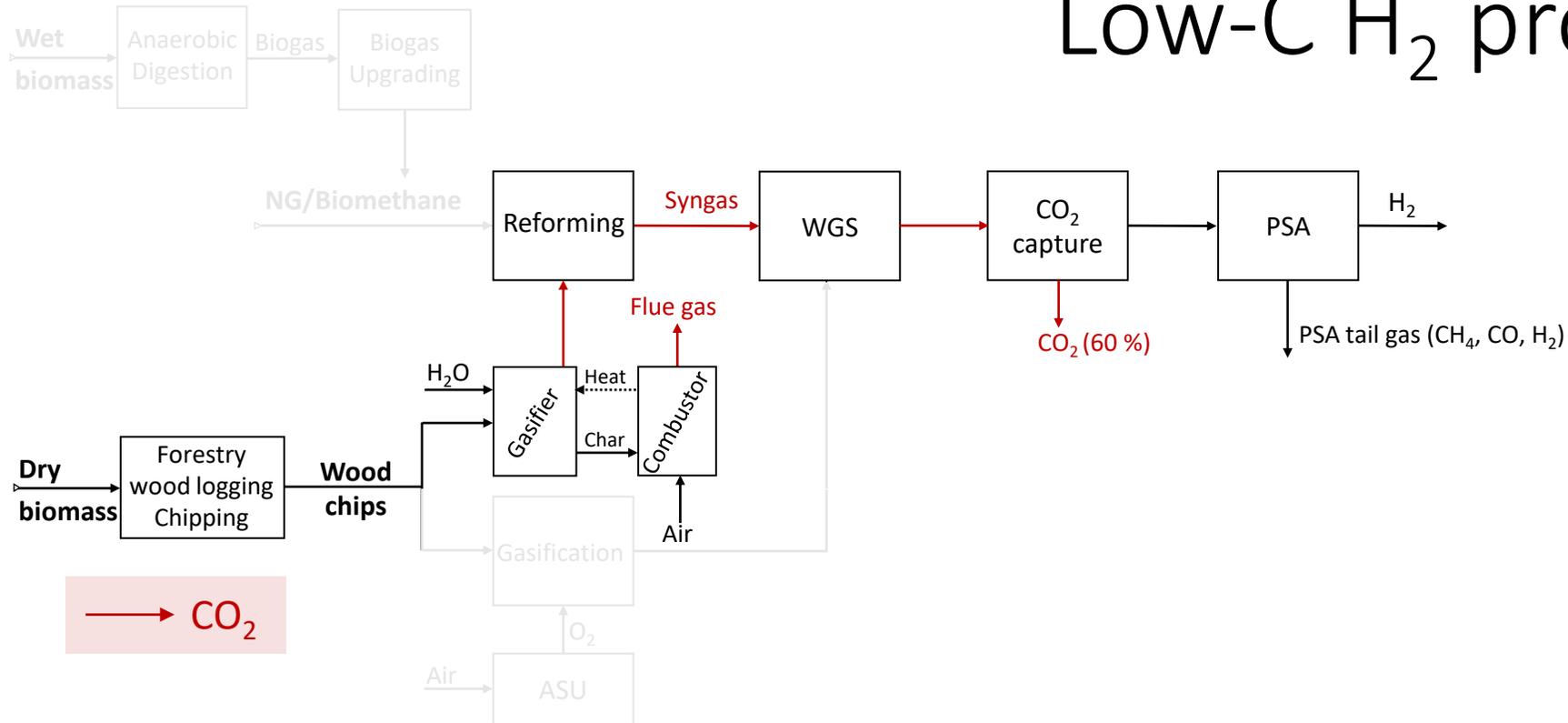
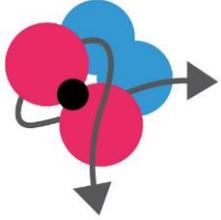
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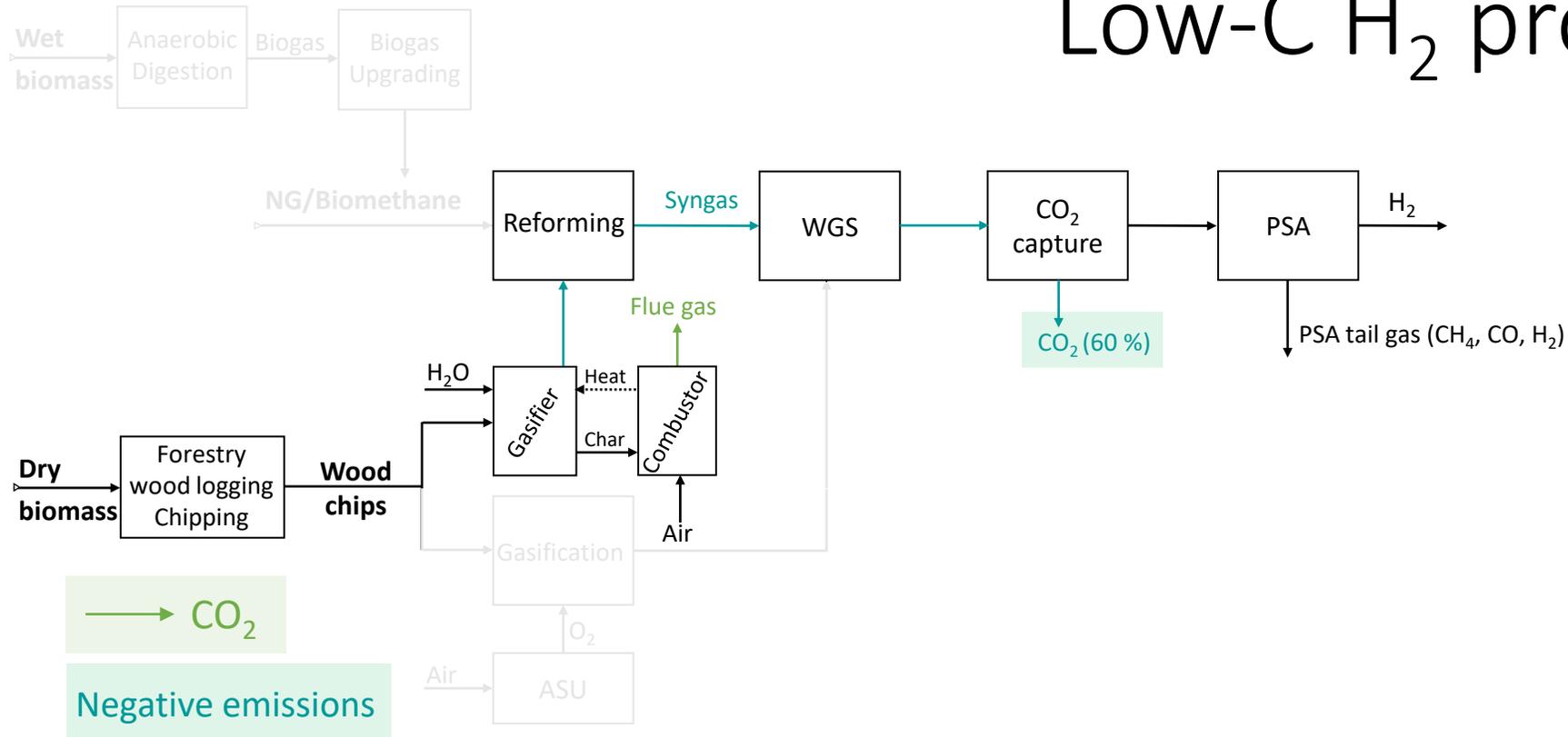
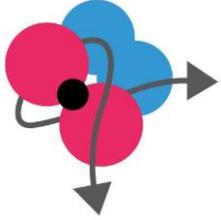


→ CO<sub>2</sub>

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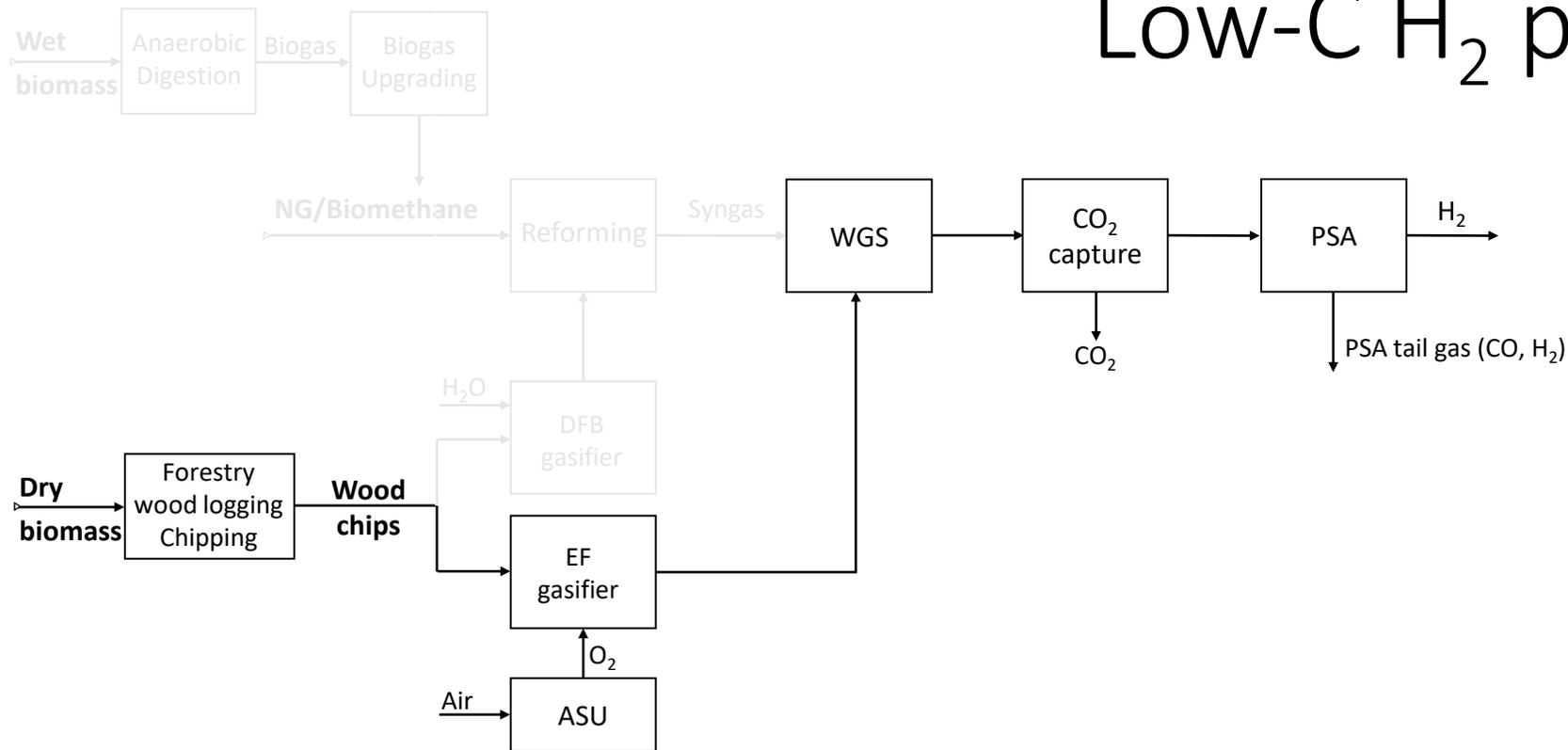
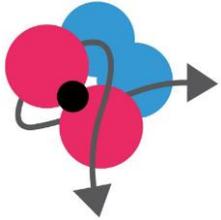
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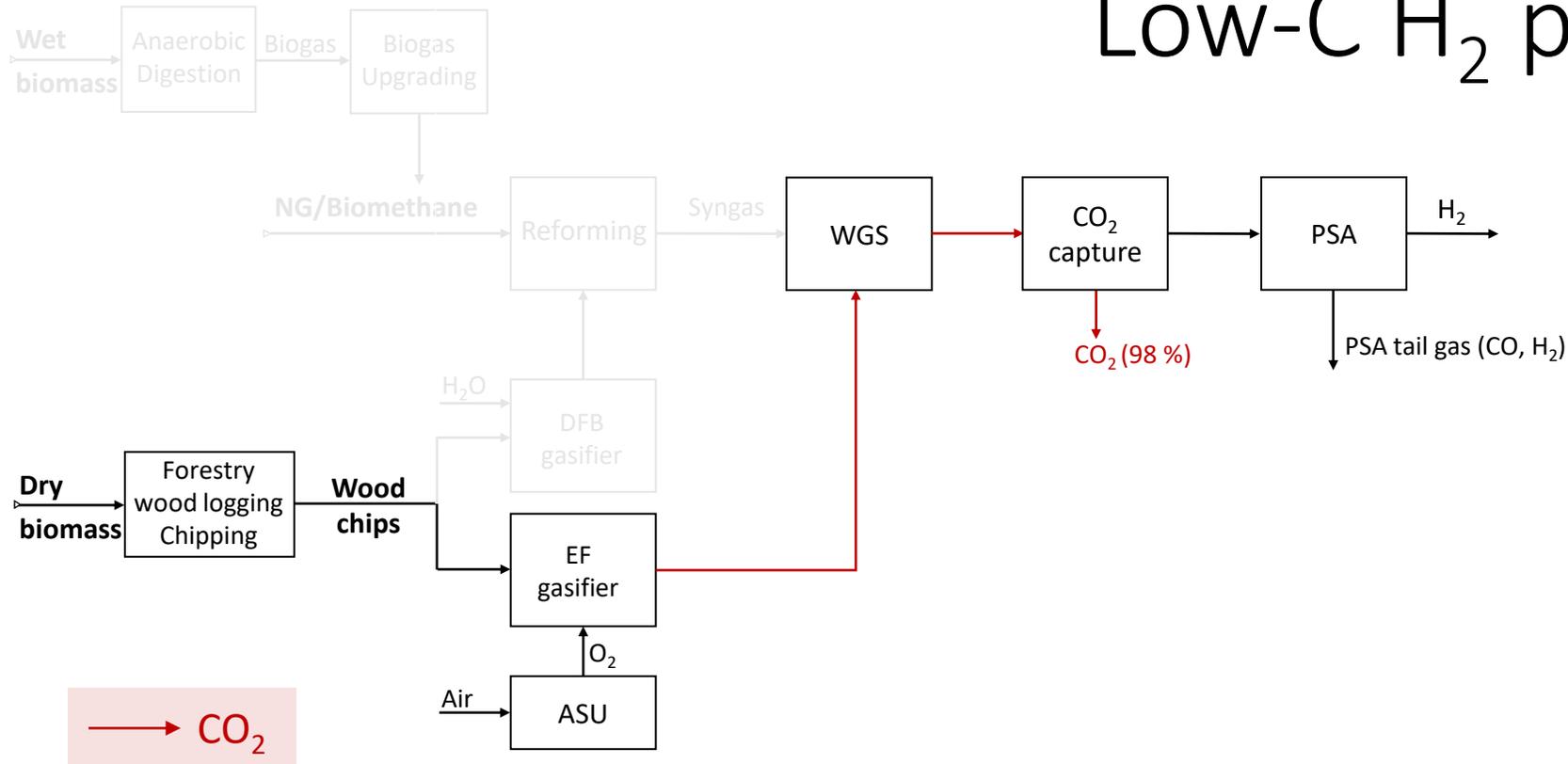
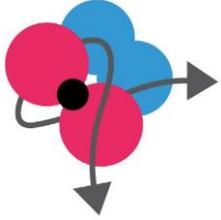
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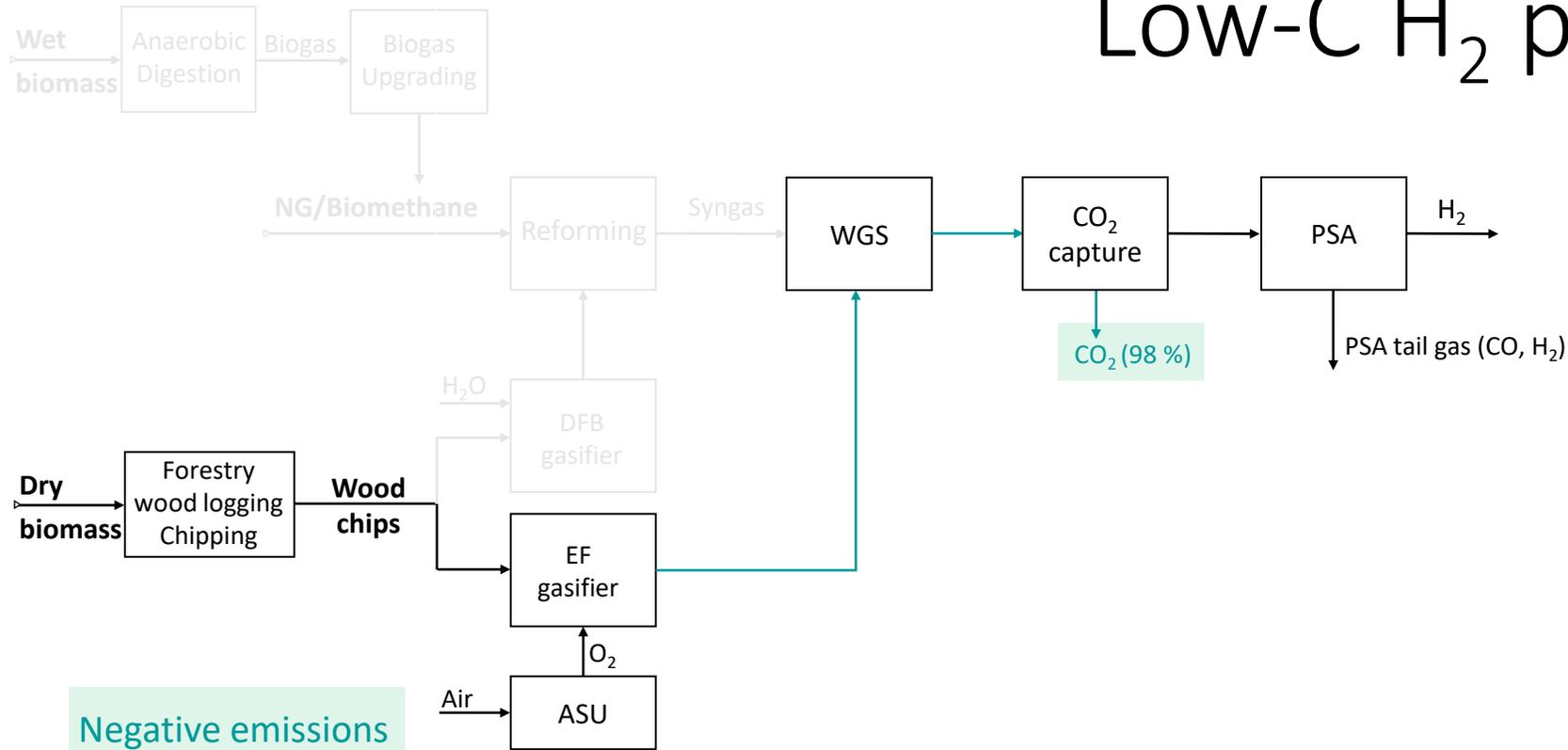
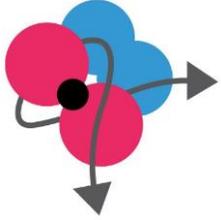
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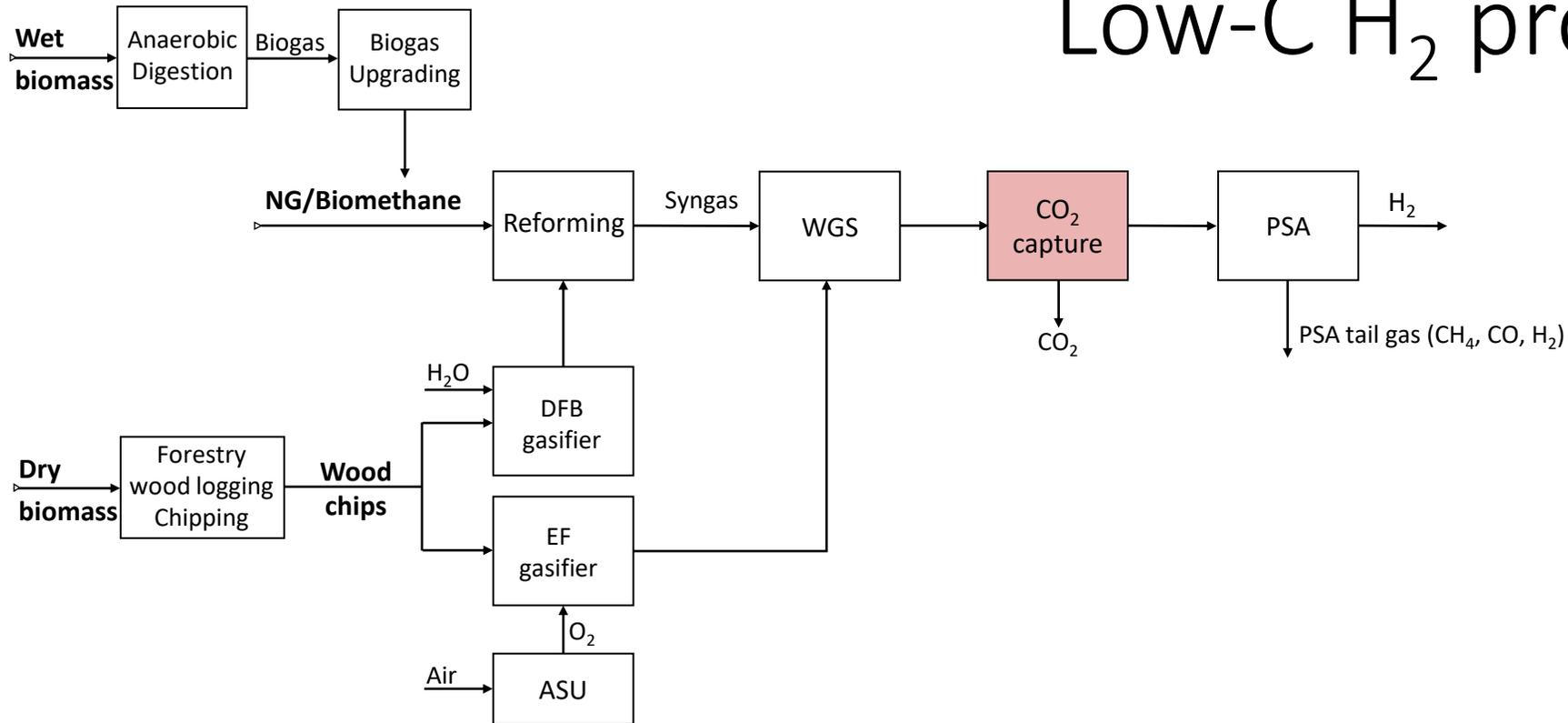
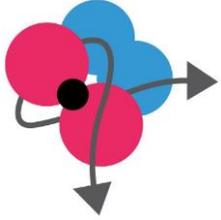


Negative emissions

Feedstock	Feedstock conversion	Technology
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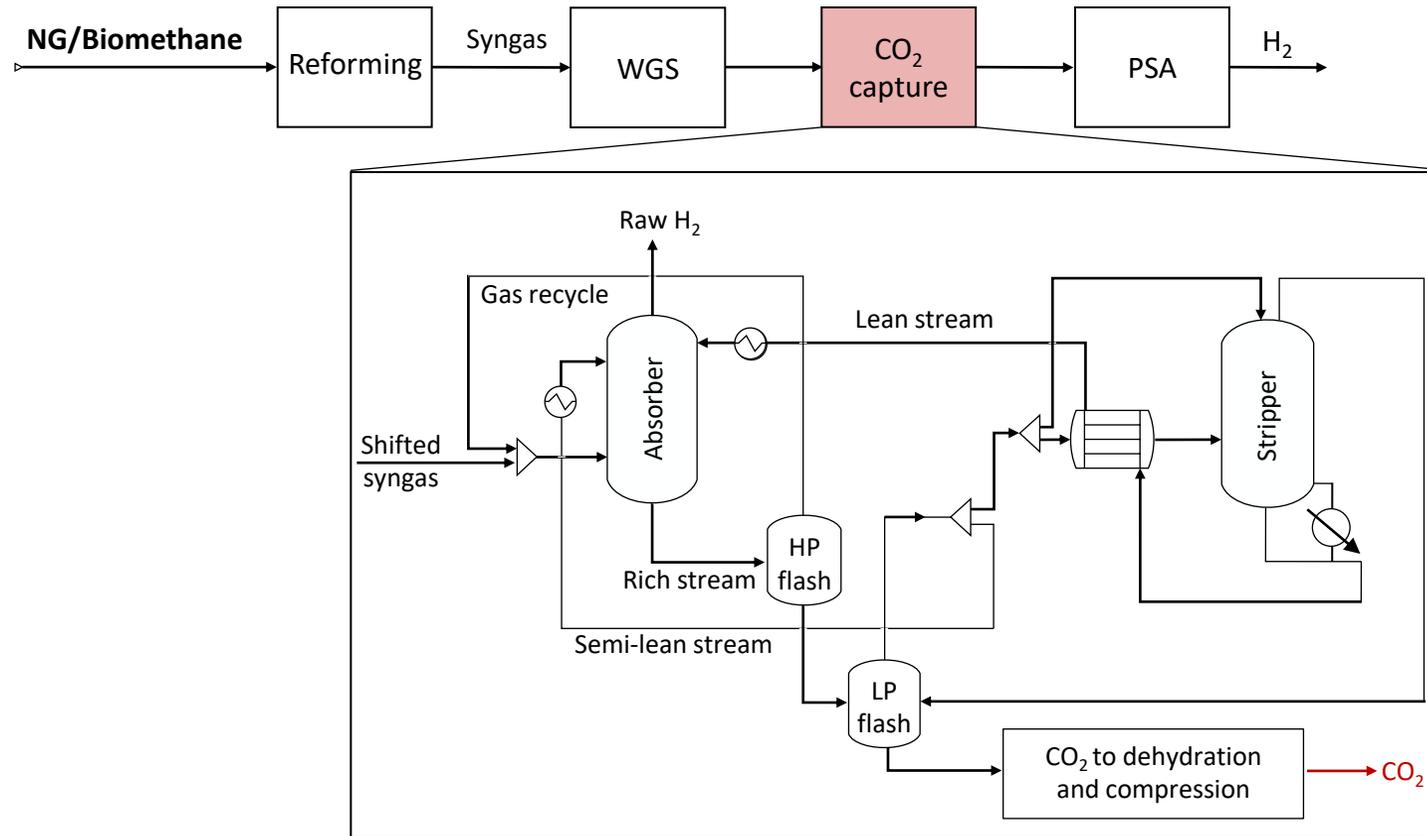
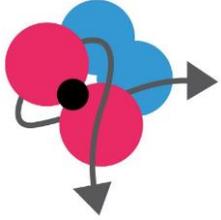


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➤ These production chains are modelled in Aspen Plus

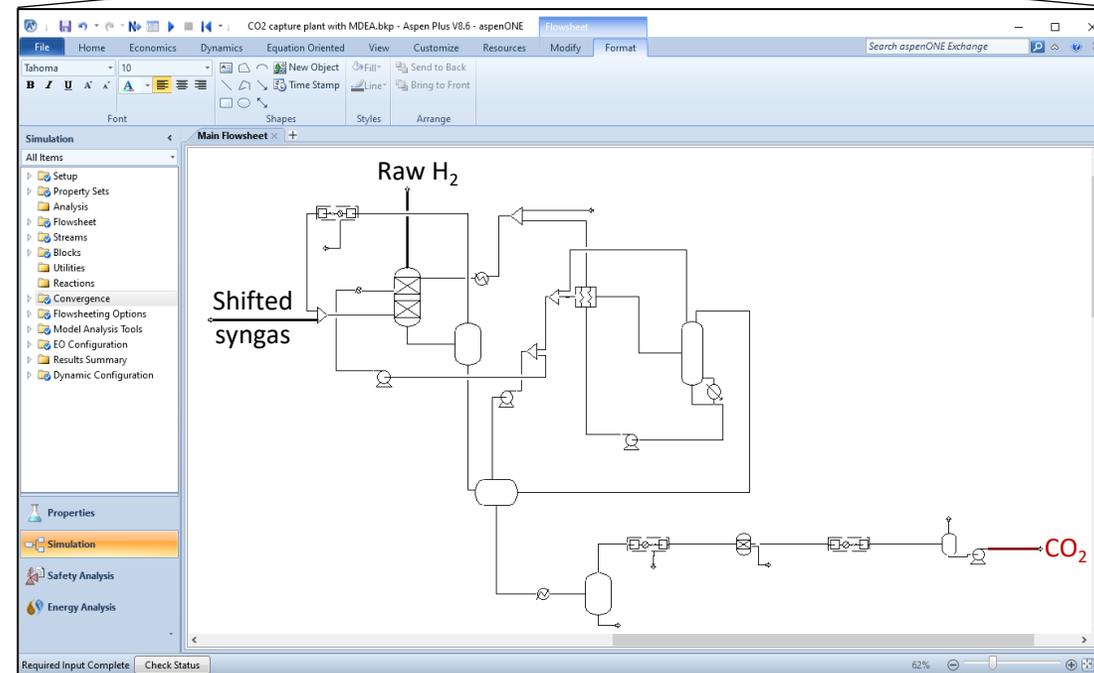
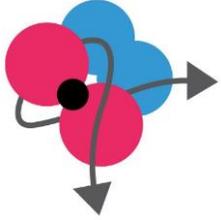
# CO<sub>2</sub> capture - benchmark technology



Benchmark: amine absorption  
 Standard solvent: Methyl diethanolamine (MDEA)

NG: Natural Gas  
 WGS: Water-gas shift section  
 (V)PSA: (Vacuum) Pressure swing adsorption  
 LP: Low pressure  
 HP: High pressure

# CO<sub>2</sub> capture - benchmark technology

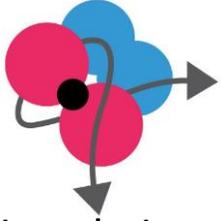


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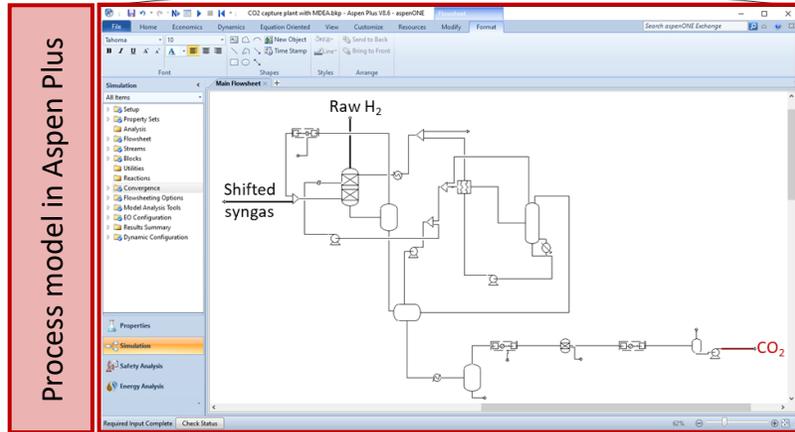
➤ Detailed model in Aspen Plus

NG: Natural Gas  
WGS: Water-gas shift section  
(V)PSA: (Vacuum) Pressure swing adsorption  
LP: Low pressure  
HP: High pressure

# CO<sub>2</sub> capture technology – optimization



Goal: optimize the CO<sub>2</sub> capture plant for different H<sub>2</sub> production chains



## ➤ Detailed model in Aspen Plus

- 1) Sensitivity analysis on the process variables
- 2) Select the decision variables, continuous ( $\mathbf{x}$ ) and discrete ( $\mathbf{y}$ )
- 3) Two objective functions to optimize
  - CO<sub>2</sub> capture rate  $\psi$
  - Total specific equivalent work  $\omega$
- 4) The optimization problem is solved using a genetic algorithm

→ Pareto Optimum

CO<sub>2</sub> capture rate:

$$\psi = \frac{m_{\text{CO}_2}^{\text{captured}}}{m_{\text{CO}_2}^{\text{in}}}$$

Total specific equivalent work:

$$\omega = \frac{W_{\text{tot}}}{m_{\text{CO}_2}^{\text{captured}}} \text{ [MJ/kgCO}_2\text{]}$$

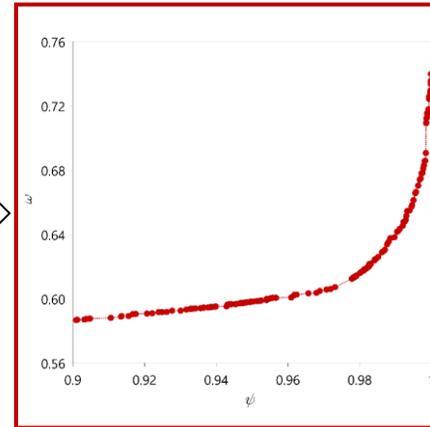
Multi-objective optimization

$$\min_{\mathbf{x}, \mathbf{y}} \left[ \omega, \frac{1}{\psi} \right]$$

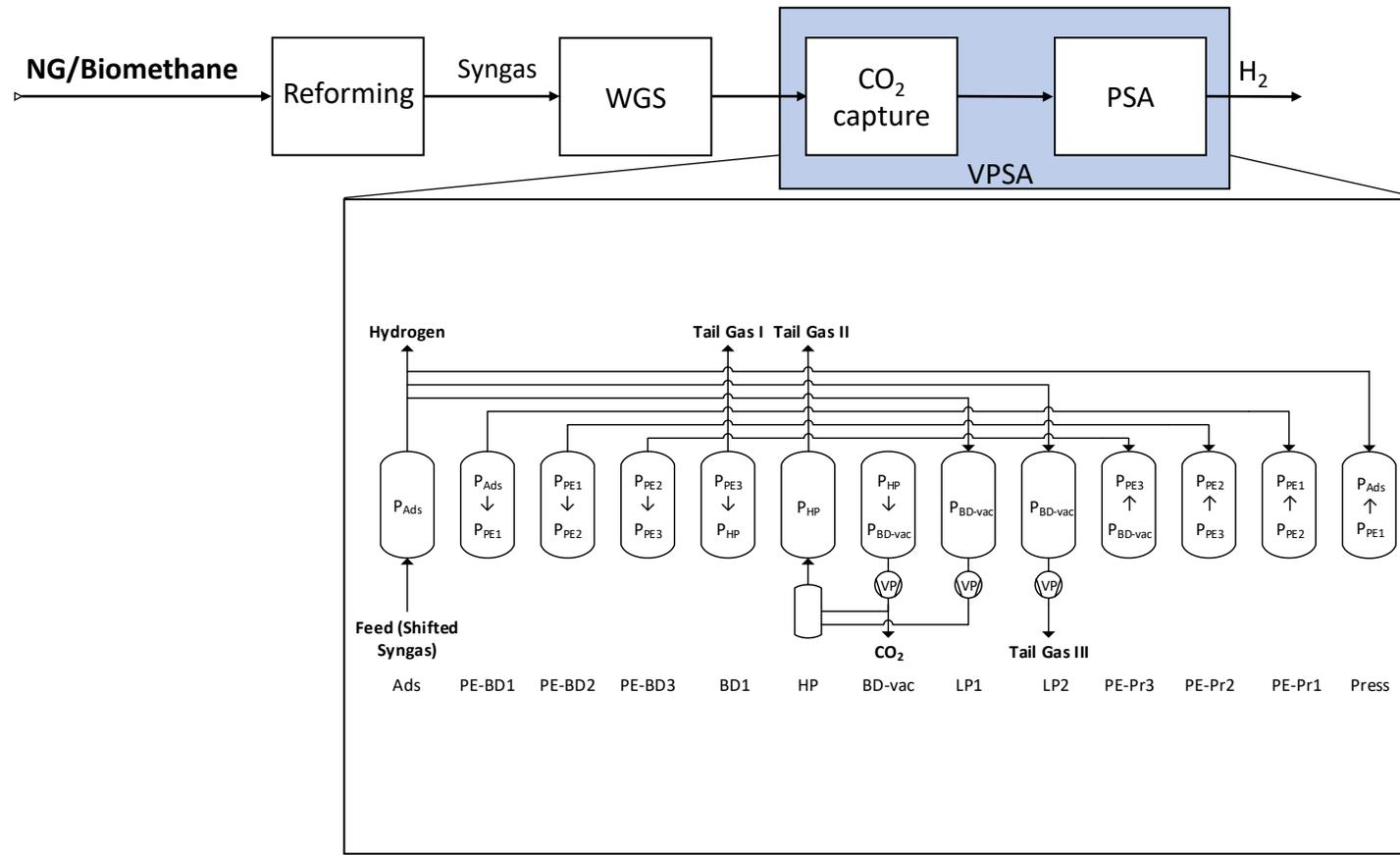
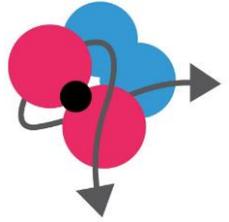
subject to

$$\mathbf{x}_{\min} \leq \mathbf{x} \leq \mathbf{x}_{\max}$$

$$\mathbf{y} \in \{\mathbf{y}_1, \mathbf{y}_2, \dots, \mathbf{y}_N\}$$



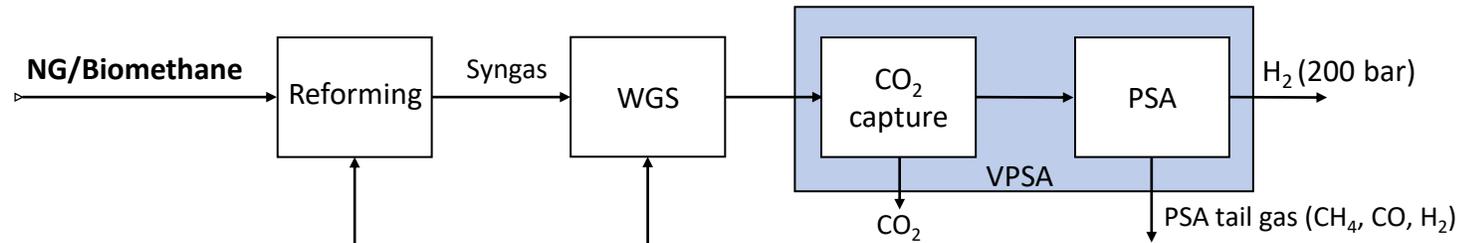
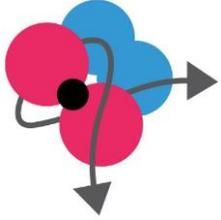
# CO<sub>2</sub> capture - novel technology



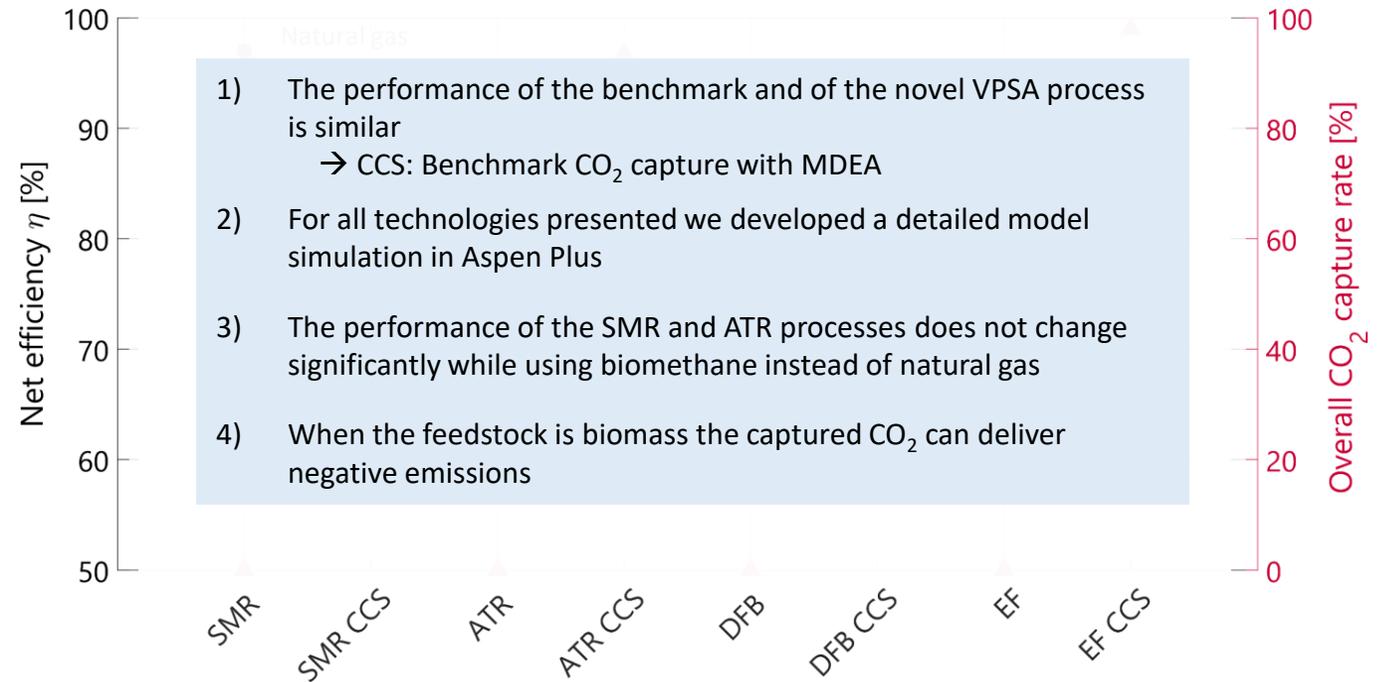
NG: Natural Gas  
 WGS: Water-gas shift section  
 (V)PSA: (Vacuum) Pressure swing adsorption

Novel CO<sub>2</sub>/H<sub>2</sub> separation technology: Vacuum pressure Swing Adsorption

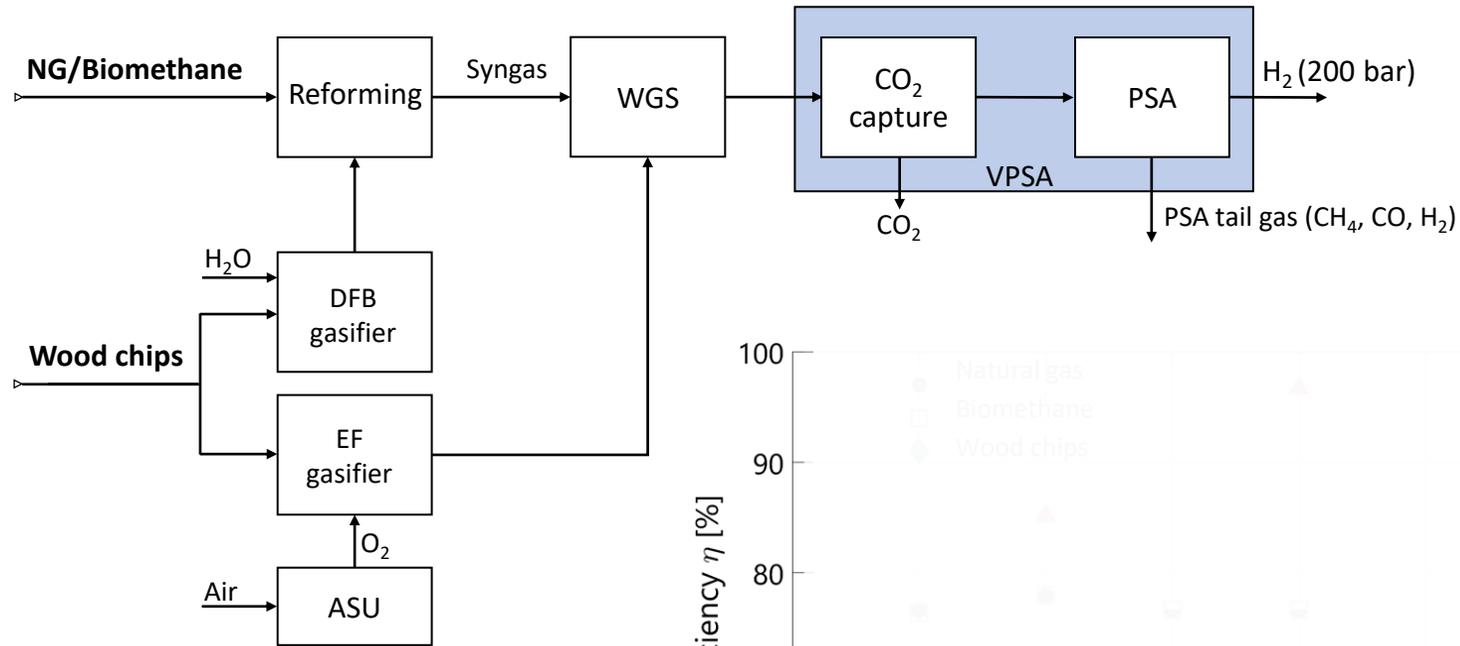
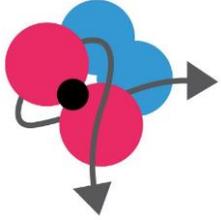
# Low-C H<sub>2</sub> production - performance



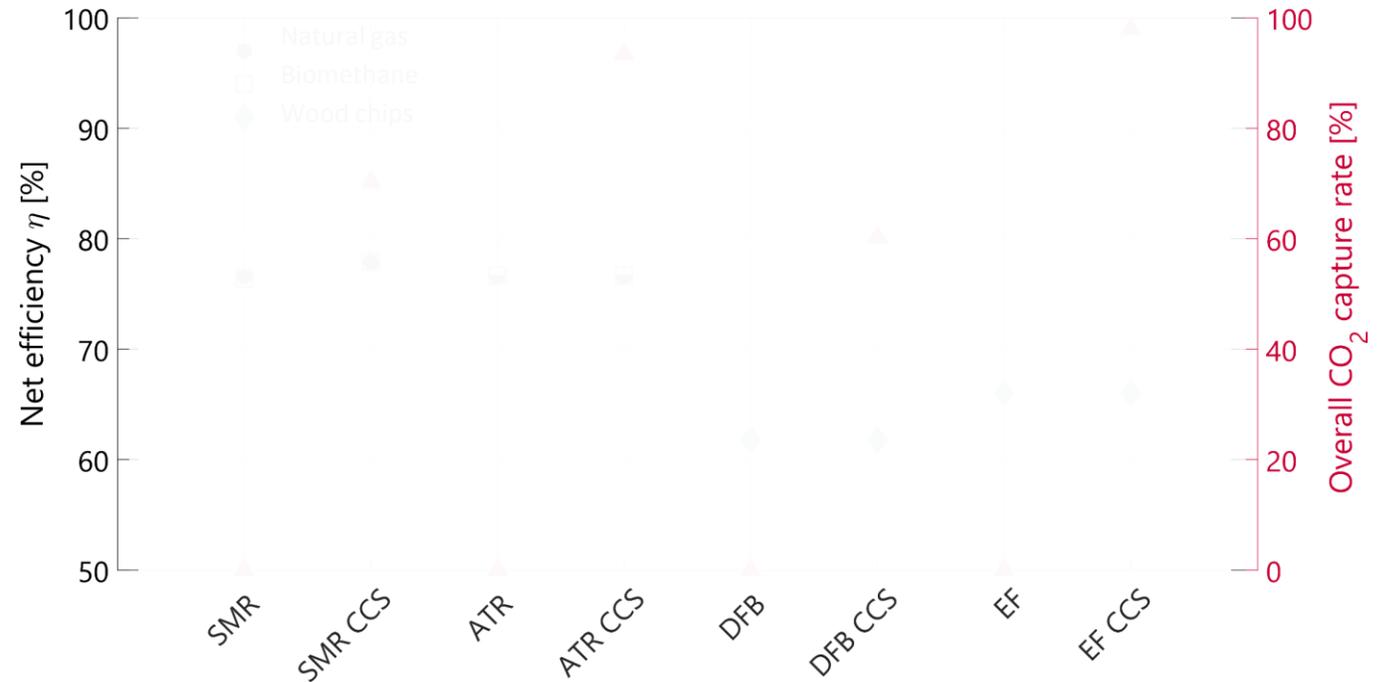
$$\eta = \frac{E_{H_2} [\text{MW}]}{E_{\text{feedstock}} [\text{MW}]}$$



# Low-C H<sub>2</sub> production - performance



$$\eta = \frac{E_{\text{H}_2} [\text{MW}]}{E_{\text{feedstock}} [\text{MW}]}$$

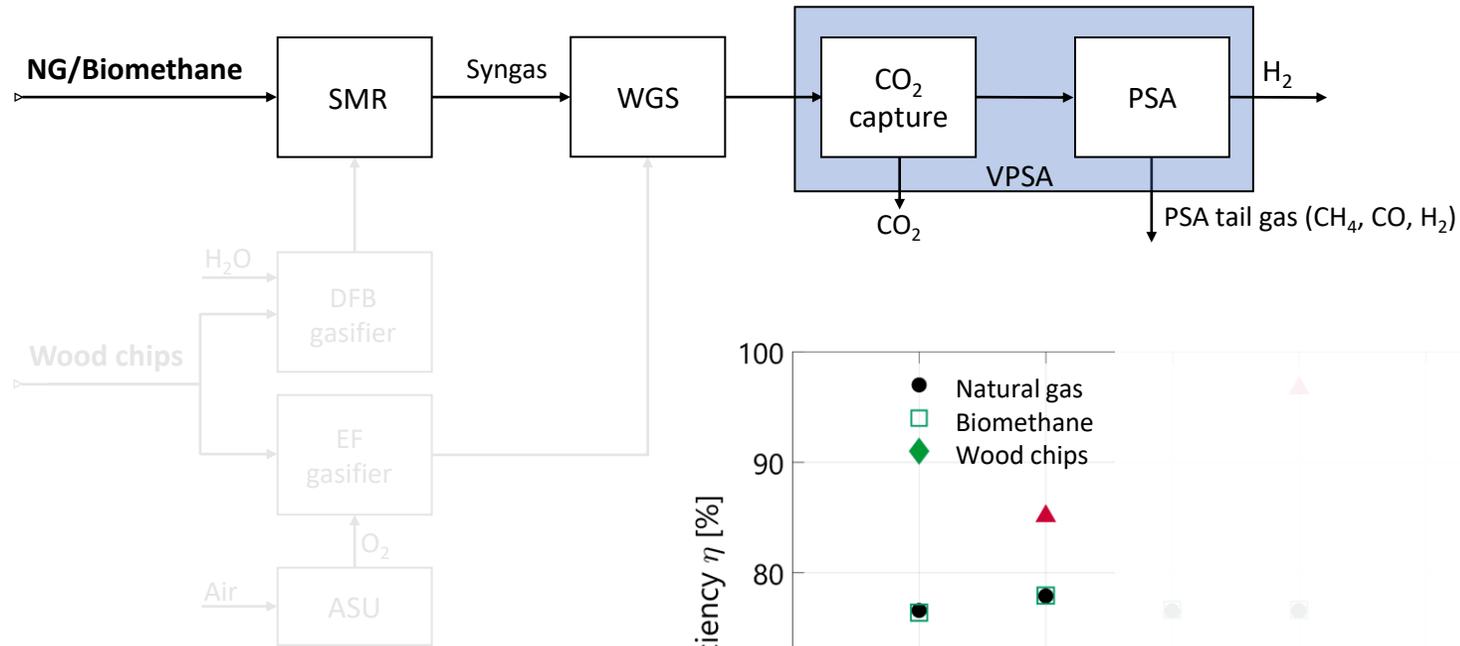
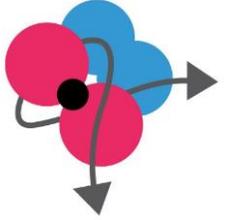


SMR: Steam methane reforming  
DFB: Dual fluidized bed

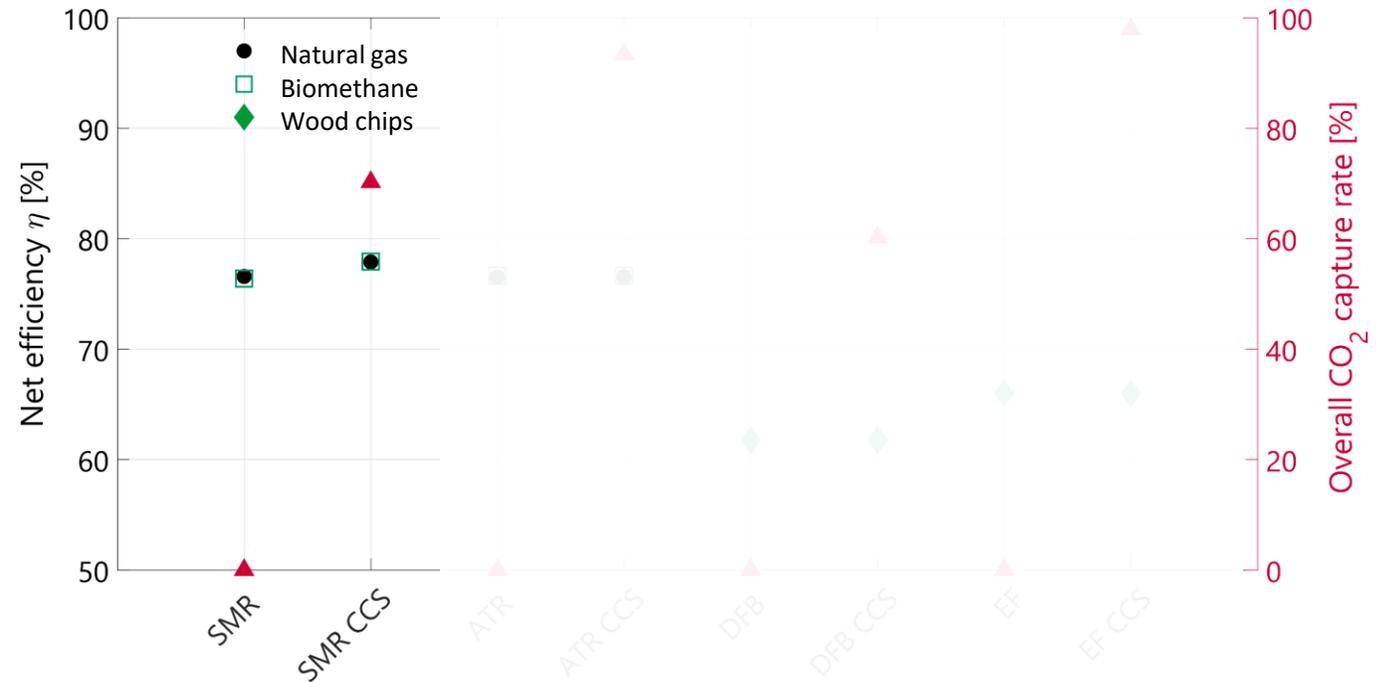
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EF: Entrained flow

CCS: benchmark CO<sub>2</sub> capture with MDEA

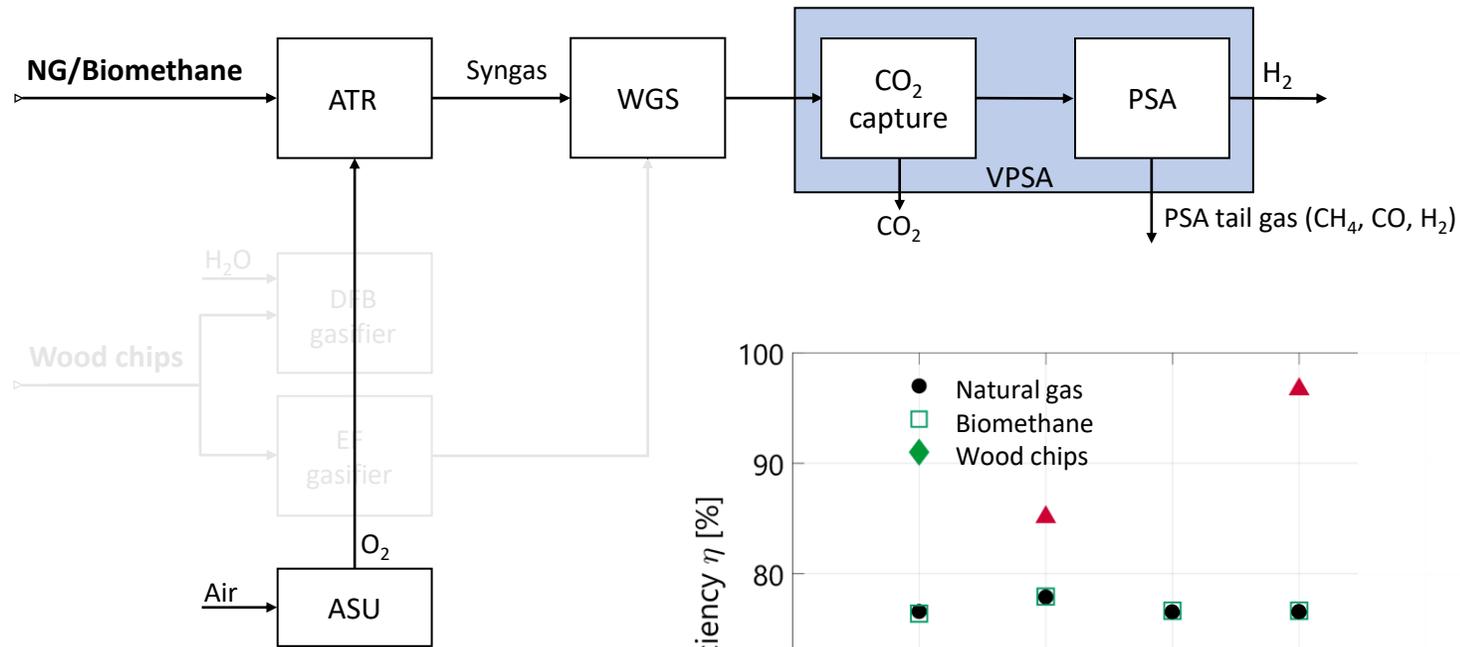
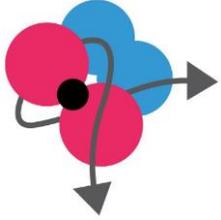
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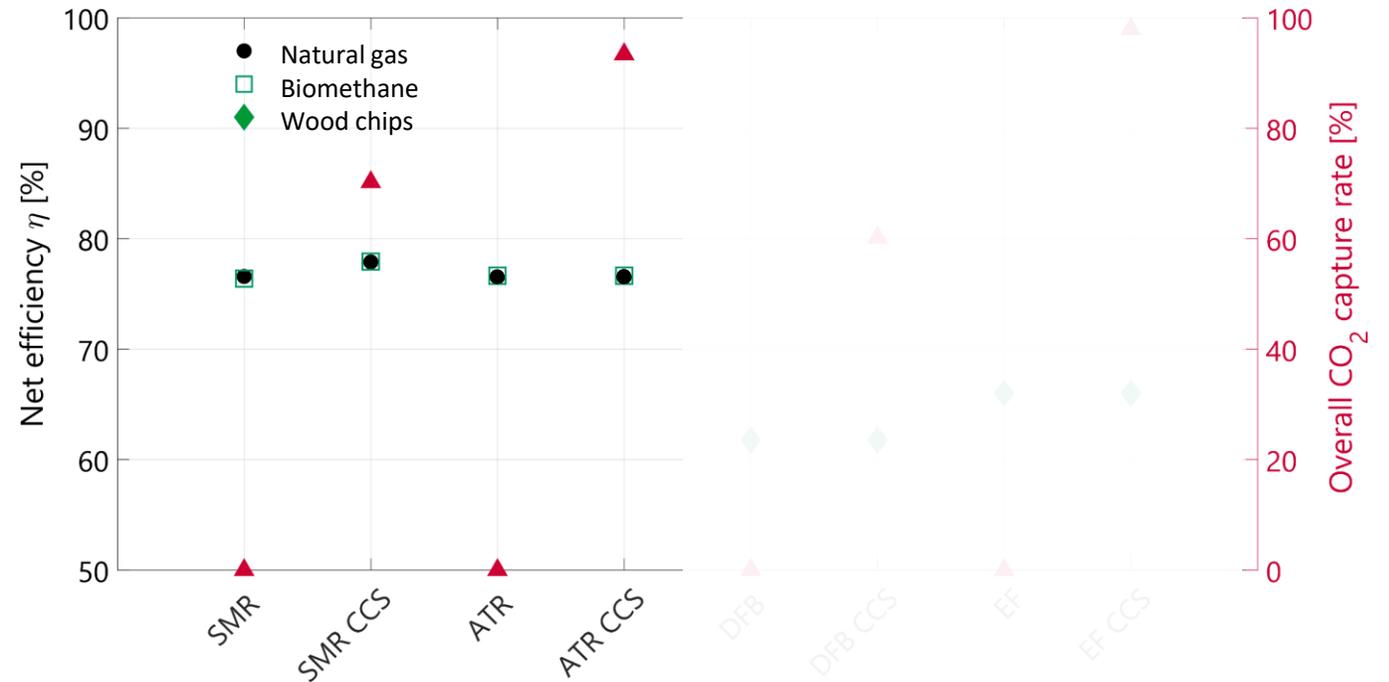
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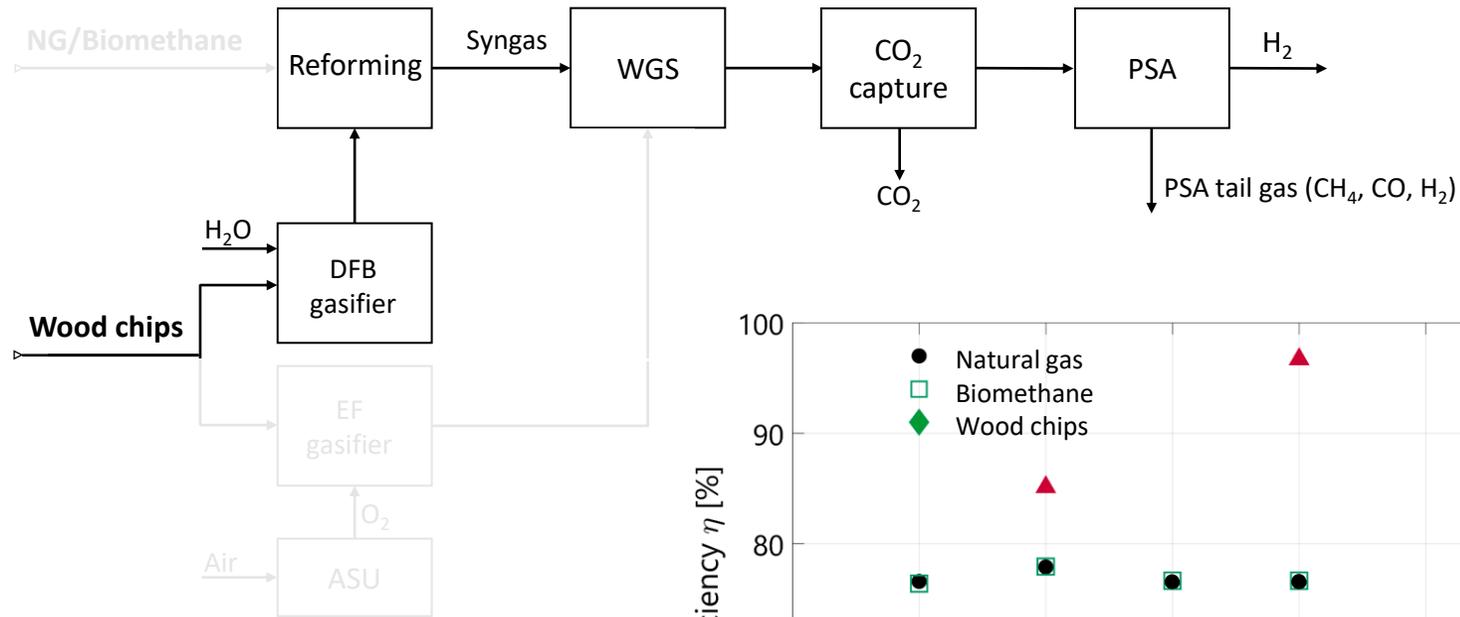
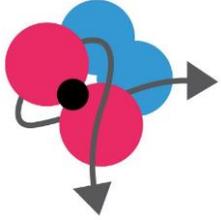
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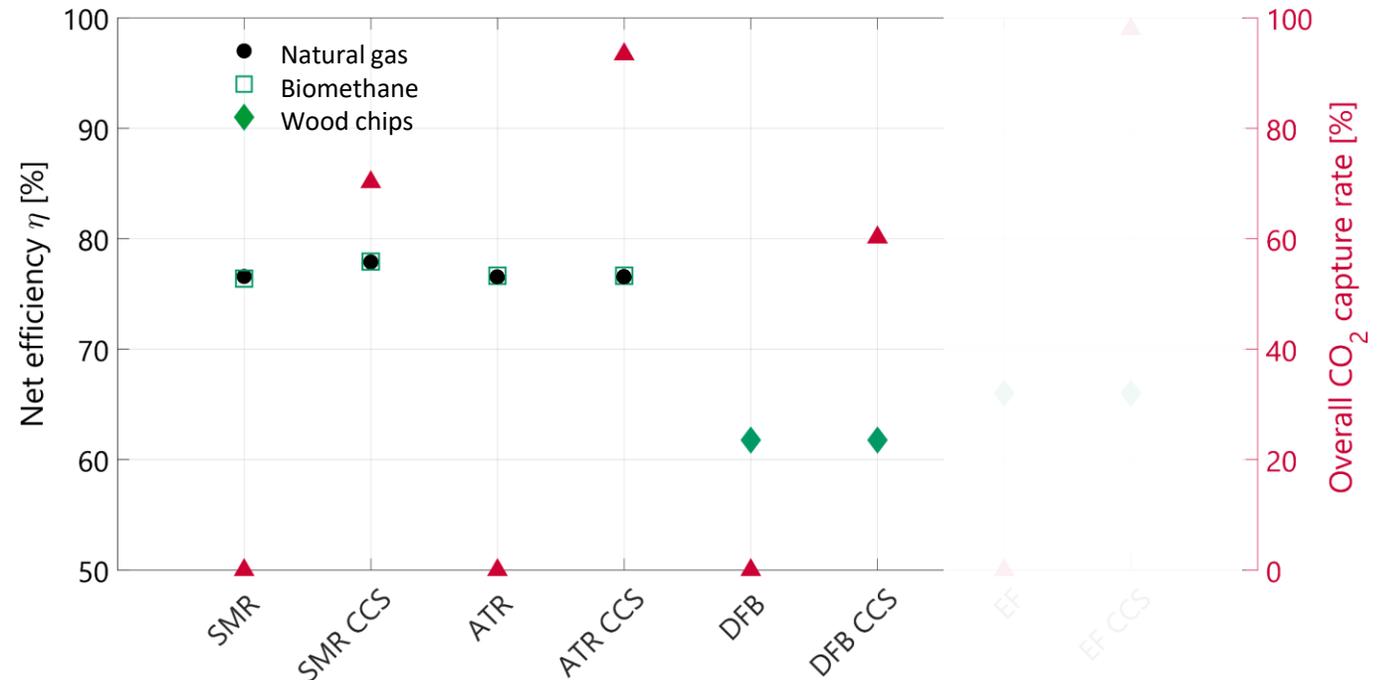
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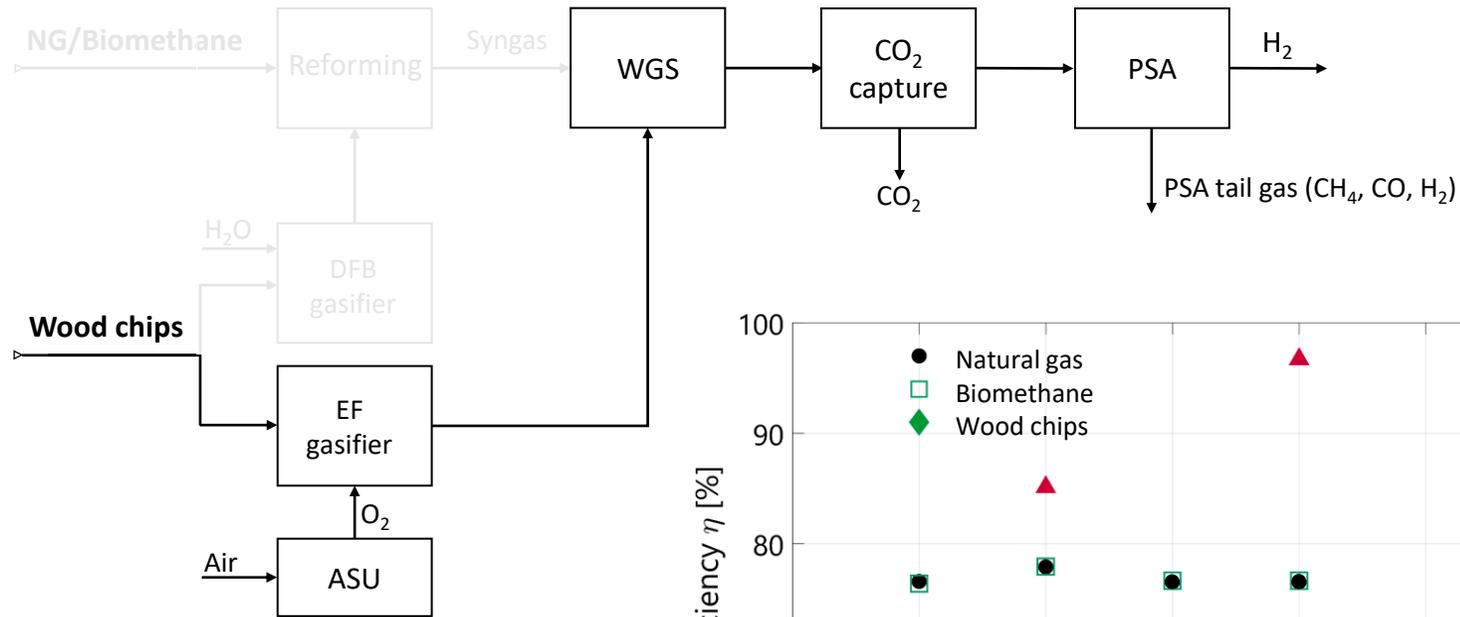
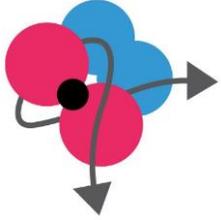


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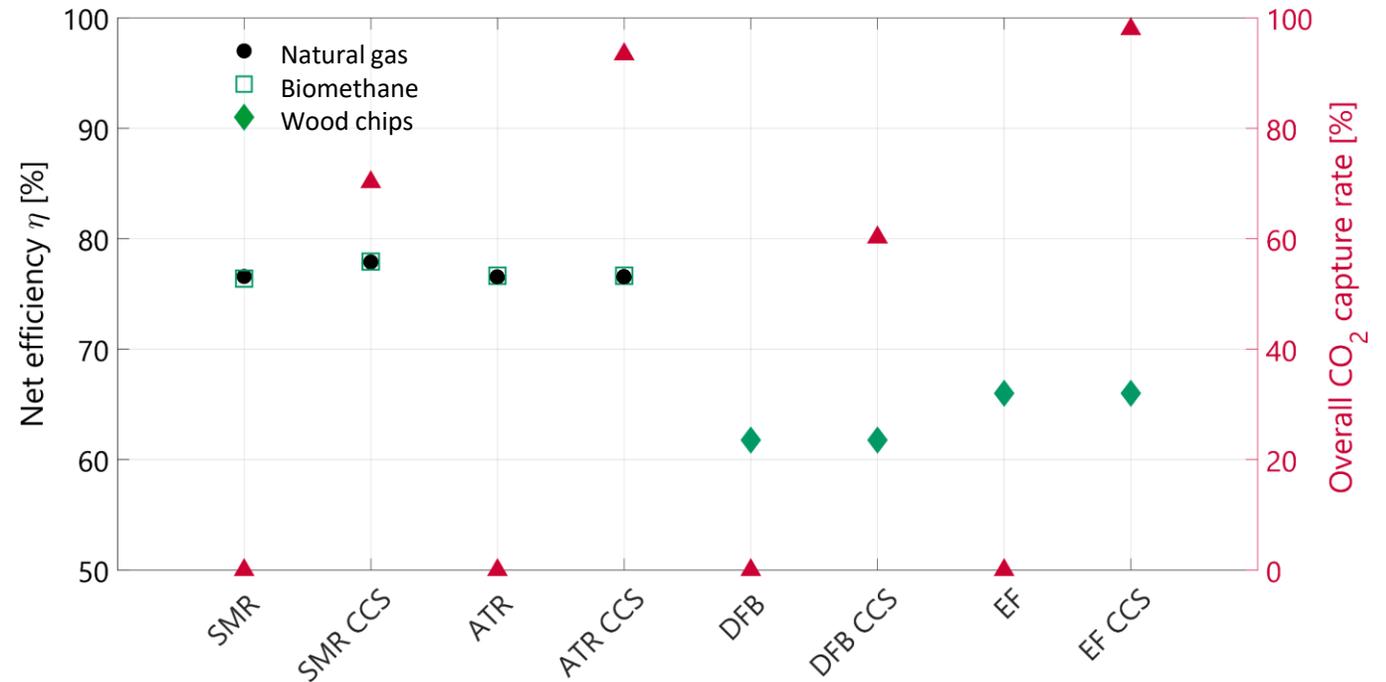
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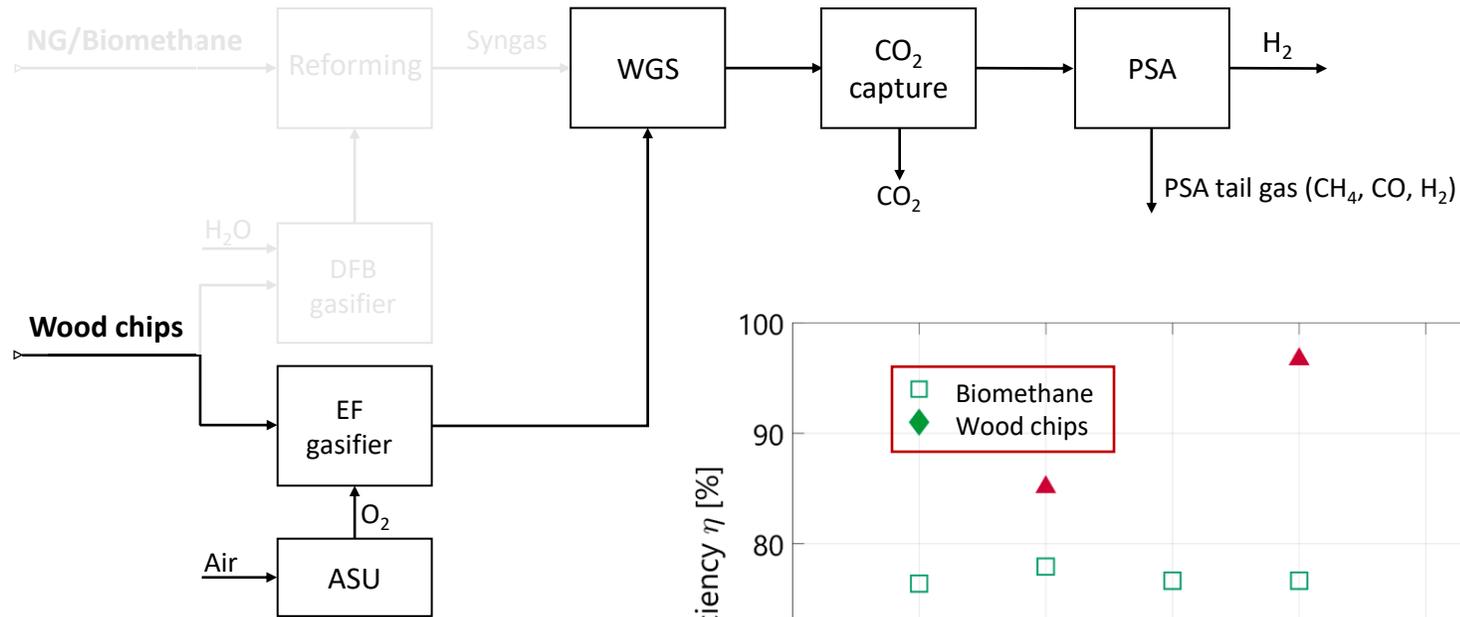
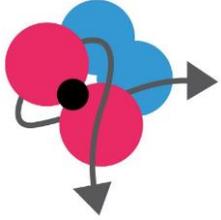


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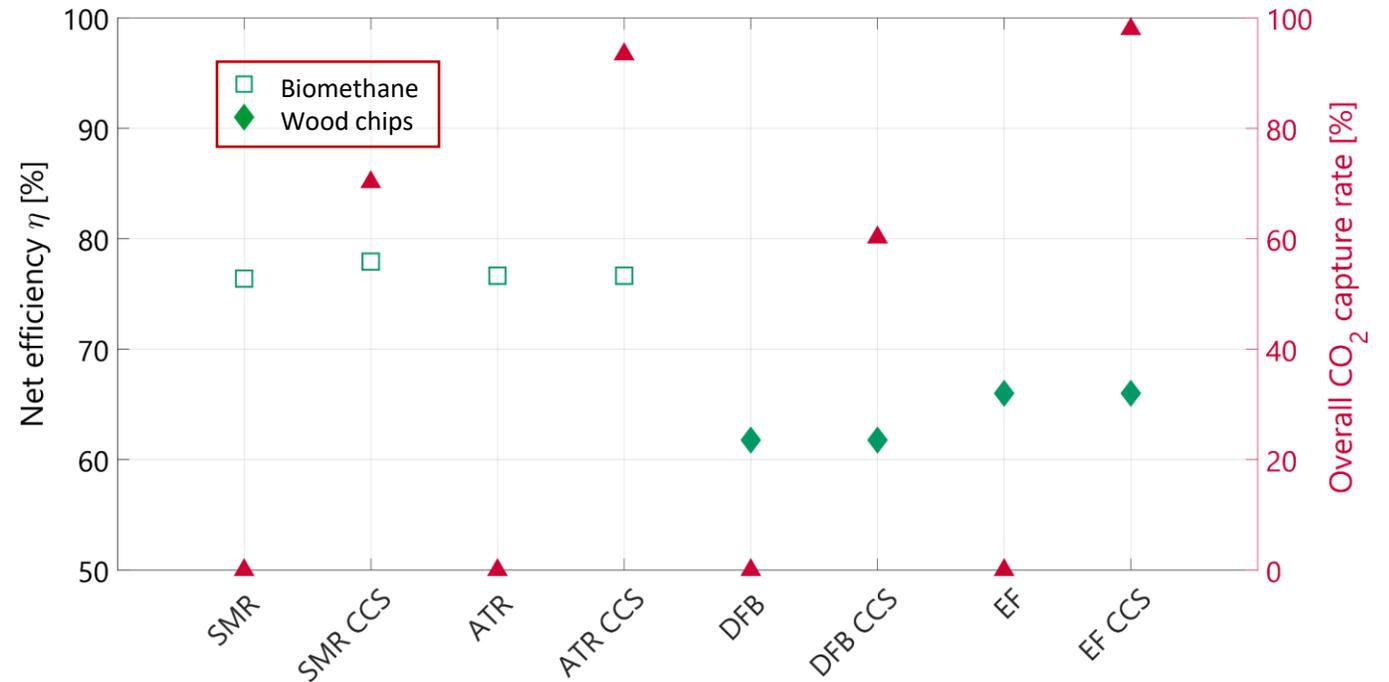
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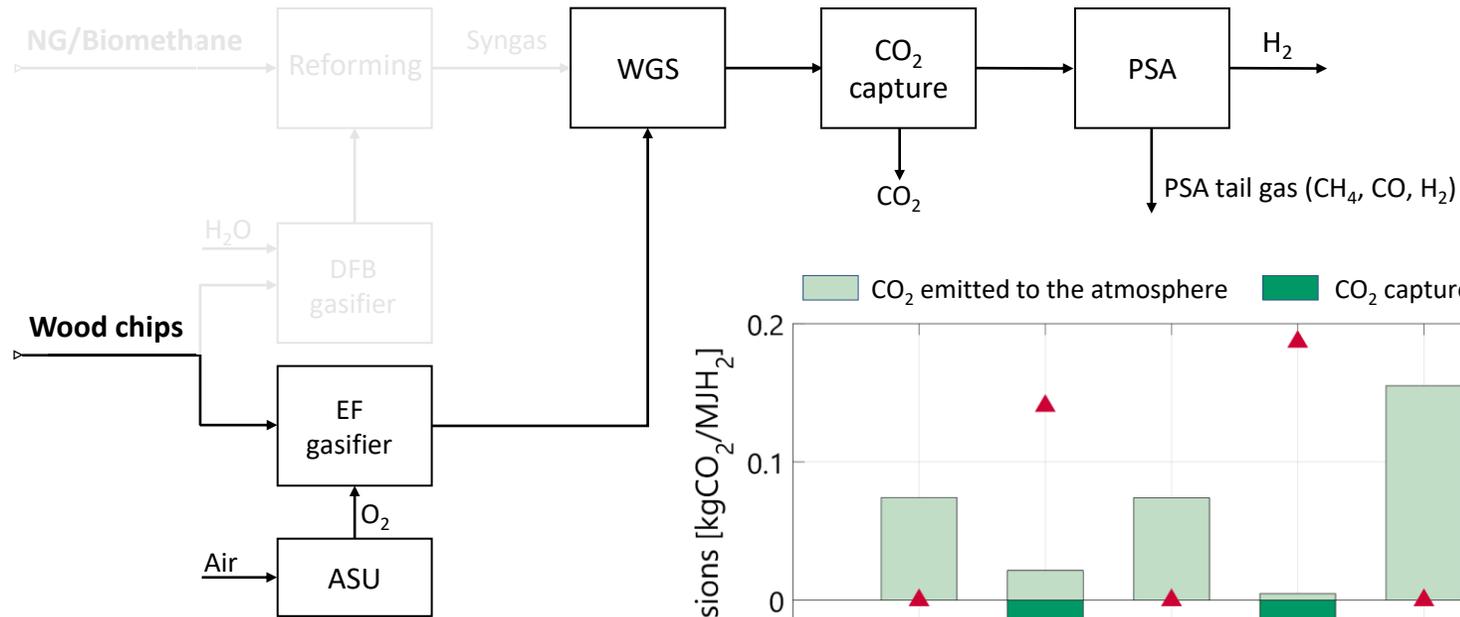
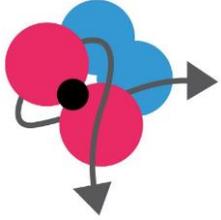
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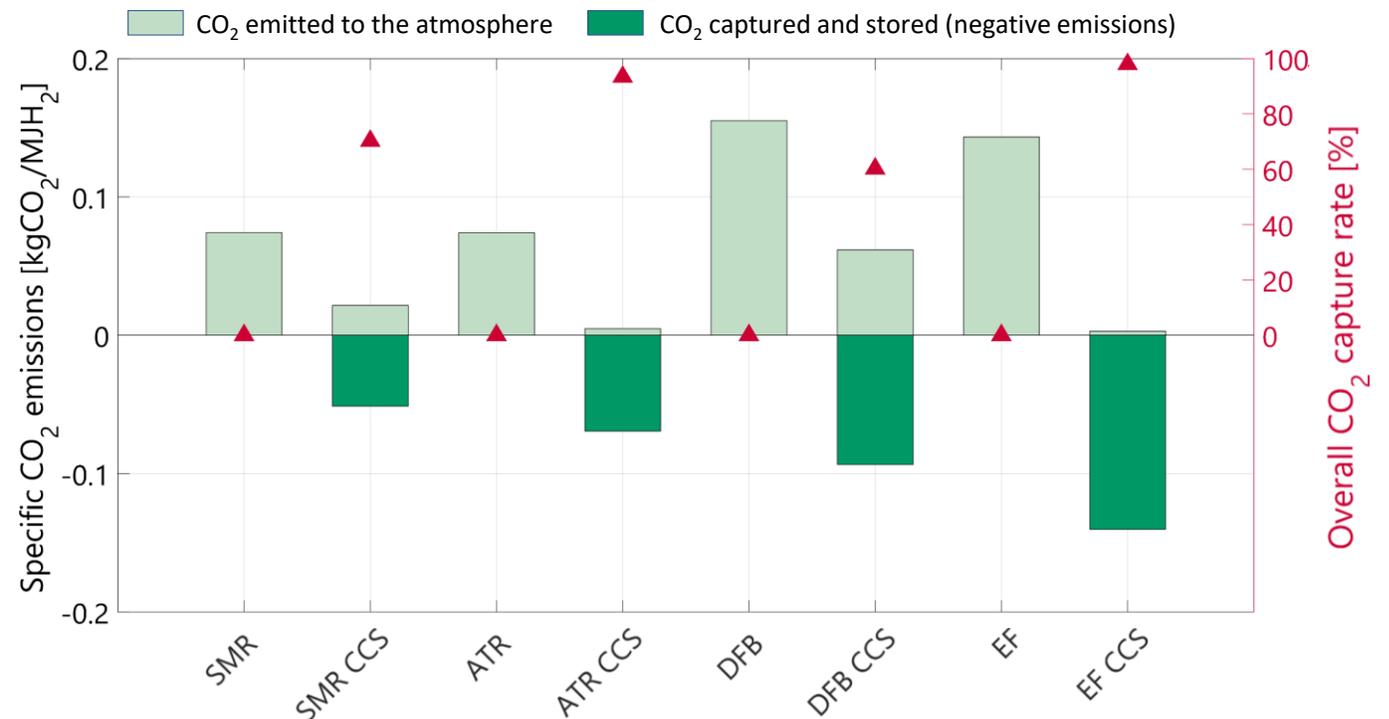
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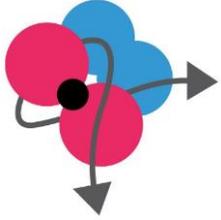
# Low-C H<sub>2</sub> production - performance



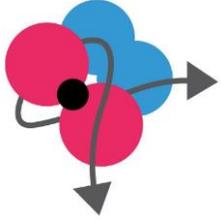
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# Conclusions



- Producing low carbon hydrogen is possible
  - Combining production from natural gas with CCS
  - Using biomass either wet or dry
  - Negative emissions can be obtained when combining hydrogen production from biomass with CCS
- The availability of biomass will be a critical factor and will vary from region to region
- The overall environmental performance that depends on the type of feedstock and production chain selected has to be evaluated via LCA
  - Karin will present the overall environmental performance of low-C hydrogen production

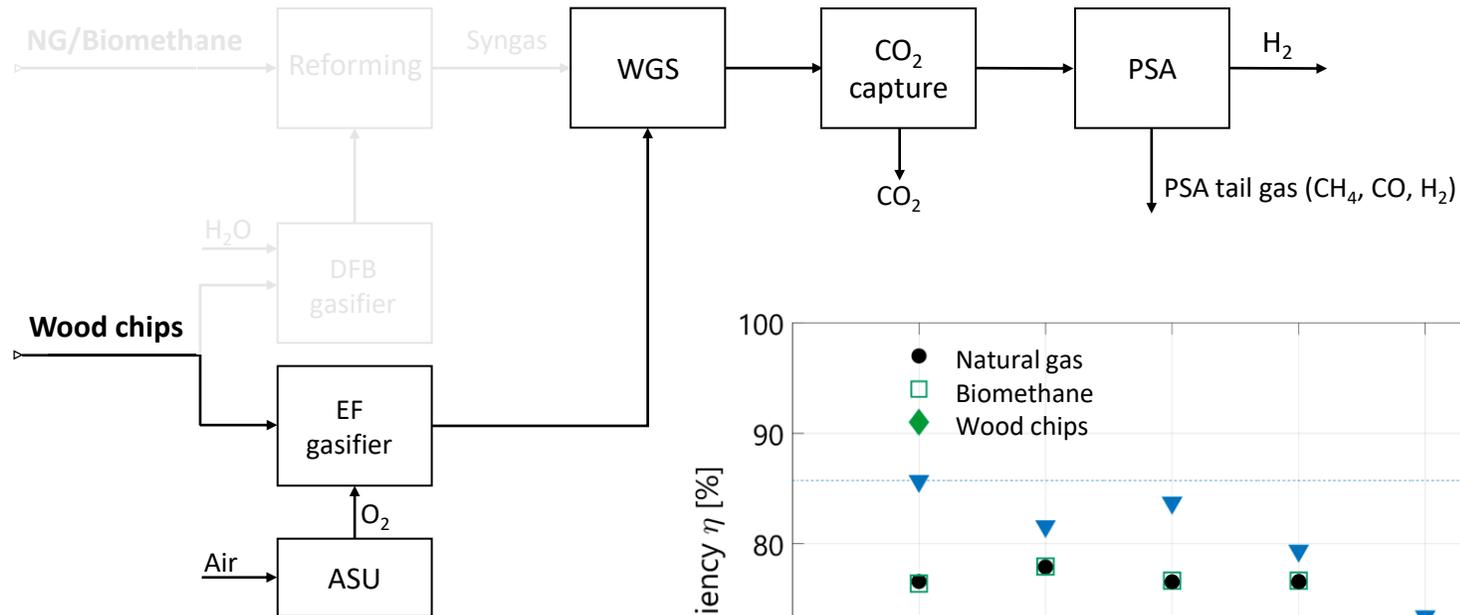
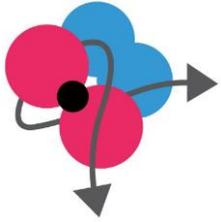


# Acknowledgement

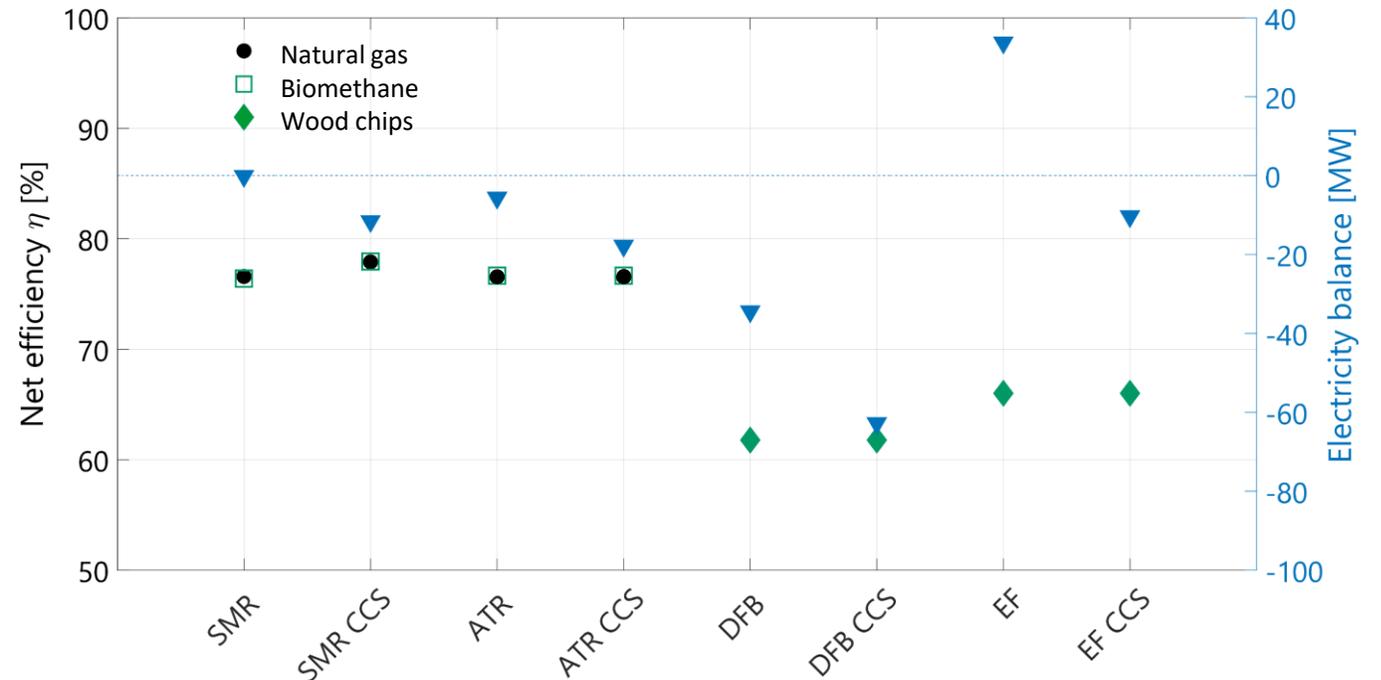
ACT ELEGANCY, Project No 271498, has received funding from DETEC (CH), BMWi (DE), RVO (NL), Gassnova (NO), BEIS (UK), Gassco, Equinor and Total, and is cofunded by the European Commission under the Horizon 2020 programme, ACT Grant Agreement No 691712. This project is supported by the pilot and demonstration programme of the Swiss Federal Office of Energy (SFOE).



# Low-C H<sub>2</sub> production - performance



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