# Development of proton ceramic electrolysers using tubular design



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# High pressure PCE

• Pressurized dry H<sub>2</sub>: 30 bars

Simpler system and lower cost of electrochemical compression

• pH<sub>2</sub> is balanced with pH<sub>2</sub>O+pO<sub>2</sub>

No need to handle high pressure hot oxygen (as for SOE)

Application at potentially intermediate T

Potentially increased lifetime and more efficient coupling with renewables: Solar, geothermal plants (electricity, steam, heat)







#### High temperature electrolysers with novel proton ceramic tubular modules (2014-2017)



#### PCE design

Tubular design to facilitate current collection

- Simpler sealing technology, lower sealing area
- Better stress distribution during transient conditions
- Module design enables to close off a tube / replace it

Mass scale processes for low cost production

- Water based extrusion
- Spray-coating and dip-coating
- Limited fabrication steps (in particular firing steps)









#### Materials: BZCY

- $BaZr_{1-x-y}Ce_{x}Y_{y}O_{3-d}$
- 10-20% Y ; 10-20% Ce BaZr<sub>1-x-y</sub>Ce<sub>x</sub>Y<sub>y</sub>O<sub>3-d</sub>
  - Ce improves sintering and gb conductivity compared to BZY O
  - Ce decreases stability compared to BZY ☺
- Grain growth increases specific grain boundary conductivity  $\bigcirc$ 
  - Not trivial to achieve large grains
- Sintering aids bring temperatures of BZY and BZCY to <1600°C  $\, \odot$ 
  - NiO, ZnO, CuO, etc.

Solid state reactive sintering with Ni sintering aid



T. Norby, **"Proton conductivity in perovskite oxides"**, in "Perovskite oxides for solid oxide fuel cells", T. Ishihara, ed., Springer, 2009, ISBN 978-0-387-77707-8.

## Manufacturing process of half-cells: pilot scale production







Automatic 40 tons extruder with capping, cutting systems and air lifted conveyor belt



Automatic spray-coater for 40 cm long sample (batch of 6 samples)





- Three processing steps
- One co-sintering step
- BaSO<sub>4</sub> instead of BaCO<sub>3</sub>
- Lower CO<sub>2</sub> emissions
- Lower cost





# Cell designs for various concepts of current collection



#### Various electrolytes: thickness 15-30 microns

BZCY72 // BZCY72-NiO



BZY10 // BZY10-NiO



BZY10 // BZCY72-NiO BZY10+2%Ce



Dense electrolyte @ 1550°C – 24h 1610°C – 6h

Porous electrolyte @ 1550°C – 24h 1610°C – 6h 1650°C – 6h 1670°C – 6h Dense electrolyte @ 1550°C – 24 h 1610°C – 6 h



# Half-cells before and after reduction



Green tube





Sintered reduced tube

#### Hg-porosimetry



#### Between 27-32 vol% porosity (with 60 vol% Ni)



# BZCY72 // BZCY72-NiO

1550°C – 24h



Grain size: Large: 5 microns Small: 2 microns

Grain growth



Grain size: 5-10 microns







#### Electrodes

- Pre-selection of materials based:
  - Chemical compatibility with BZCY (powders mixture annealing)
  - Stability in high steam content of sintered pellets (3 bars air; Steam 75%; T = 700 °C)
- Impedance spectroscopy measurements
  - Performance
  - Manufacturability (adhesion/quality of interfaces)
- Requirements of overall integration of cells/seals in module

- LSM:  $La_{0.8}Sr_{0.2}MnO_{3-\delta}$
- LNO:  $La_2NiO_{4+\delta}$
- LSC:  $La_{0.87}Sr_{0.13}CrO_{3-\delta}$
- BSCF:  $Ba_{0.5}Sr_{0.5}Co_{0.8}Fe_{0.2}O_{3-\delta}$
- LSCF:  $La_{0.6}Sr_{0.4}Co_{0.2}Fe_{0.8}O_{3-\delta}$  BGLC:  $Ba_{1-x}La_{x}Gd_{0.8}La_{0.2}Co_{2}O_{6-\delta}$





#### Symmetrical cells: LSM/BCZY + catalysts

*Current collector: screen-printed Au grid Electrolyte: 14 mm diameter Electrode: 9 mm diameter*  **Conditions:** Total P= 3 bar Steam 75%



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#### Symmetric cells: BGLC





# Manufacturing

Tape-test



- Manufacturing based on dipcoating/annealing using powders produced by Marion Technologies
- Dip-coating and drying in air
- Annealing in air







#### Complete cells

➤ Testing in Probostat<sup>TM</sup>

- Capped and sealed towards alumina riser
  - Sealing technology developed by CTMS
- In-situ reduction





Cell Area:

5- 11 cm<sup>2</sup>



#### BZCY - BGLC







#### Hydrogen production

| Cell | Anode     | Current<br>collector | Anode comp |
|------|-----------|----------------------|------------|
| 2    | BGLC-BZCY | Pt                   | x = 0.5    |

**Conditions:** Total P= 3 bar Cell Area: 11 cm<sup>2</sup>





| Cell | Anode     | Current<br>collector | Anode comp |
|------|-----------|----------------------|------------|
| 2    | BGLC-BZCY | Pt                   | x = 0.5    |



Effects of steam pressure







## Development of 1kW reactor at CSIC

#### Working conditions:

- ➢ Temperature: 700 °C
- Pressure:
  - Total: 20 bar
  - Steam: 5 bar
- Steam temperature: 250-300°C

#### Geometry design:

- Isothermal conditions (700°C along tubes)
- Good fluid dynamics (oxygen/steam concentration)
- Mechanical strength
- Thermal resistance
- Costs optimization (materials, manufacturing, time)

#### Sealing purpose:

- Possibility of close and turn individual tube off during operation
- Low temperature gaskets







Next steps

Game changer in high temperature steam electrolysis with novel tubular cells integrated in a 10 kW module for pressurized hydrogen production





- CoorsTek Membrane Sciences AS (Norway)
- Agencia Estatal Consejo Superior de Investigaciones Cientificas (Spain)
- CRI EHF (Iceland)
- University of Oslo (Norway)
- MC2 Ingenieria y Sistemas SL (Spain)
- Shell Global Solutions International BV (The Netherlands)







#### **GAMER** activities

- Optimisation of cell design and key enabling technologies (seals, interconnects, manifolds)
- Industrial pilot production of tubular cells
- Design and engineering of a pressurized 10 kW electrolyser
- Installation, commissioning and testing of the electrolyser
- Process design, LCA and techno-economic evaluation of the electrolyser integrated in CO<sub>2</sub> to liquid fuels/chemicals plant
- Dissemination and exploitation

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